

Vodikov „ion“

6 atomov vodika, 1 e^- ;

val. f. e^- je vsota šestih val. f.

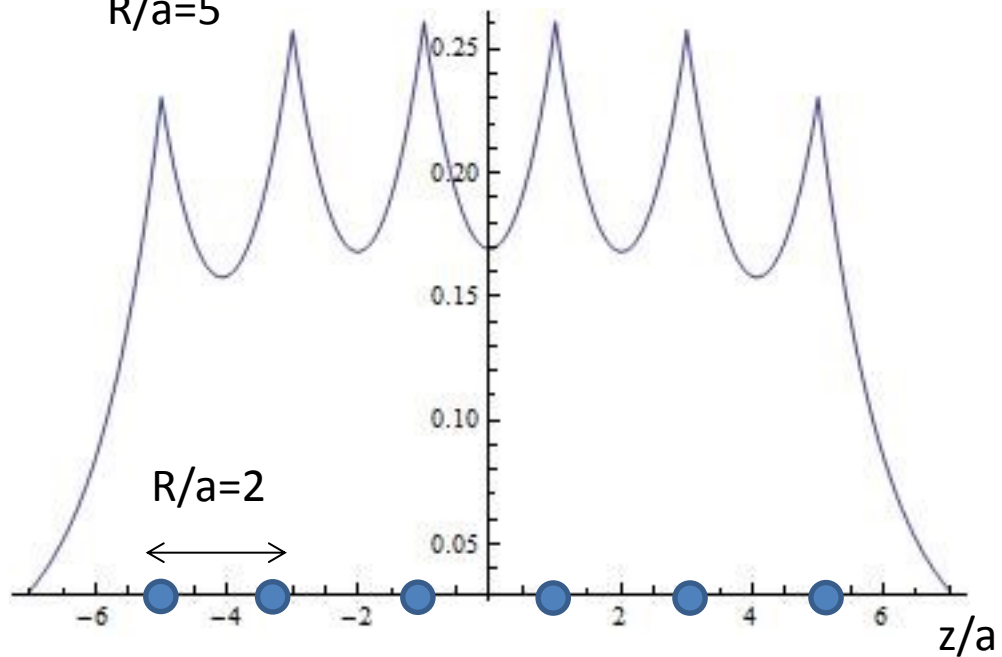
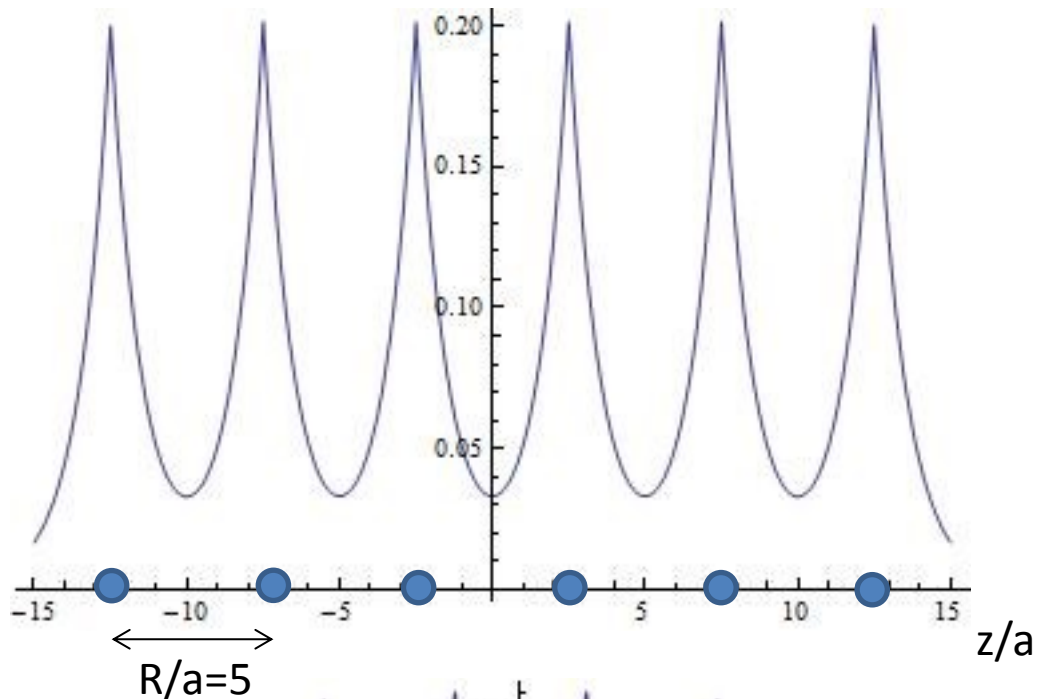
vodikovega atoma (osnovno stanje)

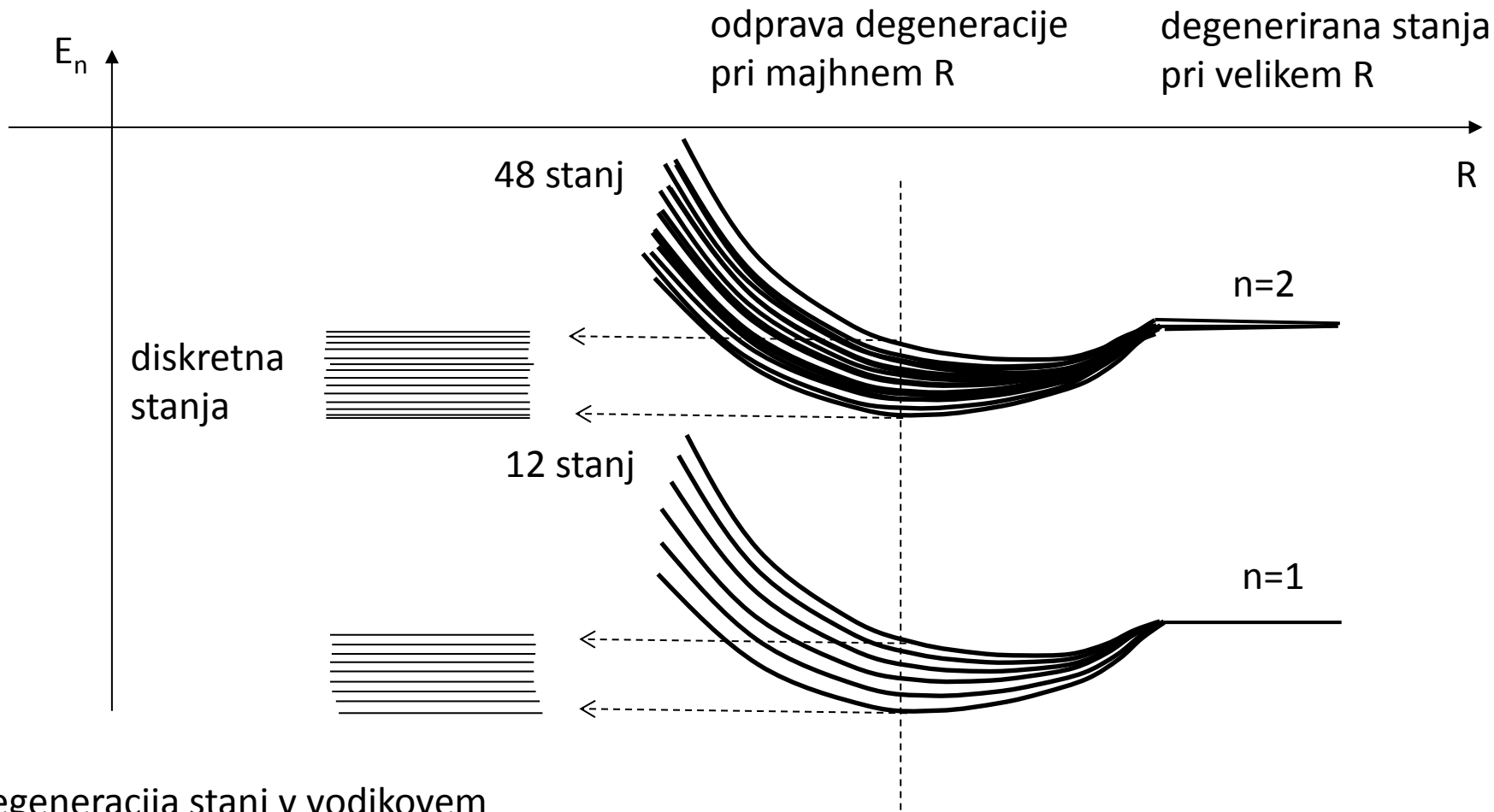
$$\psi_1(z) = \frac{1}{\sqrt{8\pi a^3}} e^{-|z|/a}$$

$$a = \frac{\pi}{\alpha c \mu}$$

$$E_n = -\frac{1}{2} c^2 \mu \frac{Z\alpha^2}{n^2}$$

pri manjšem R se stanja prekrivajo,
posledica je odprava degeneracije





degeneracija stanj v vodikovem atomu:

$$n = n_r + l + 1$$

$$l = 0, 1, \dots, n-1$$

$$m_l = -l, -l+1, \dots, l-1, l$$

število stanj z enako energijo

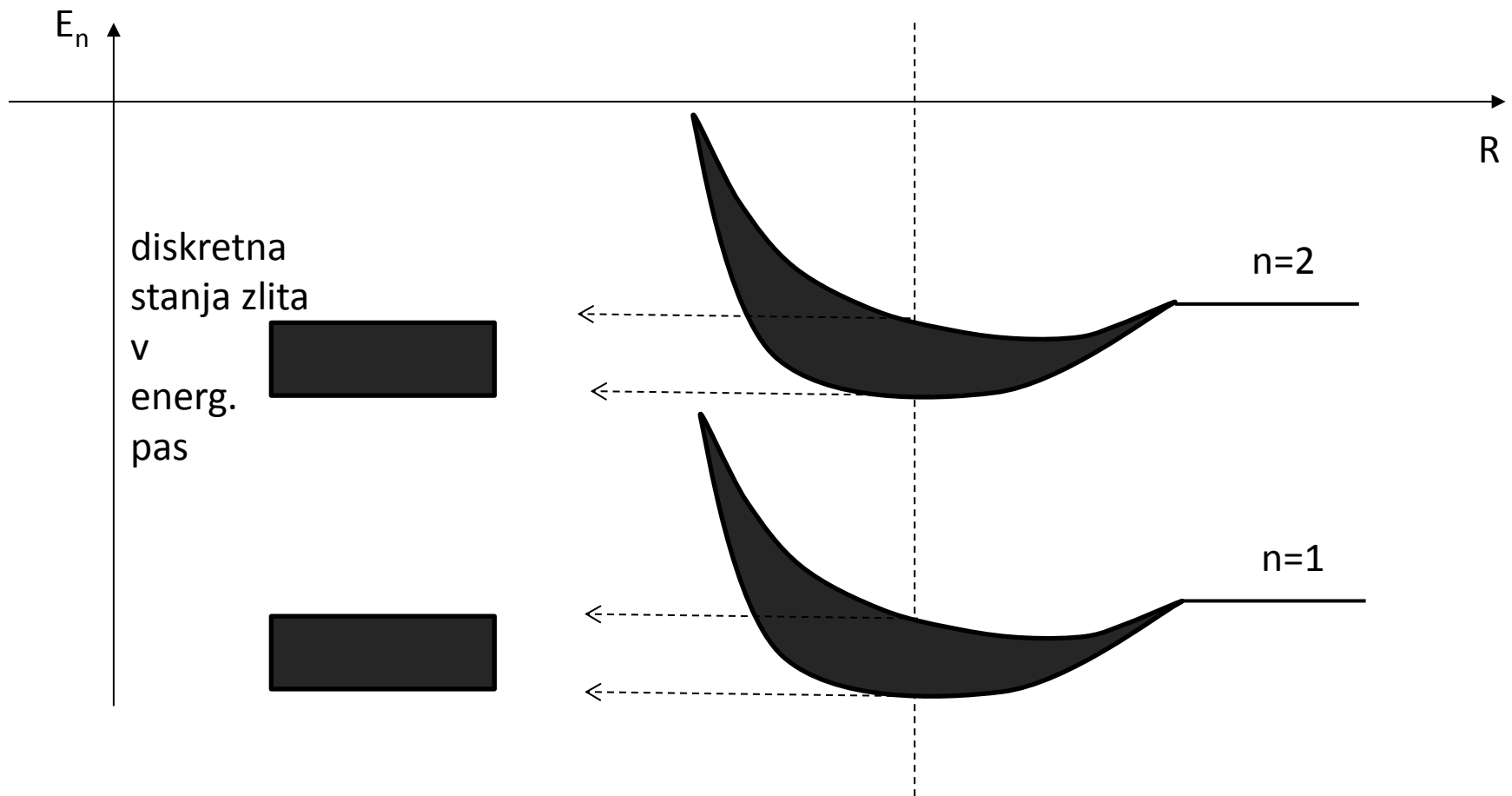
$$n^2 \rightarrow 2n^2 \text{ (spin)}$$



6 vodikovih atomov:

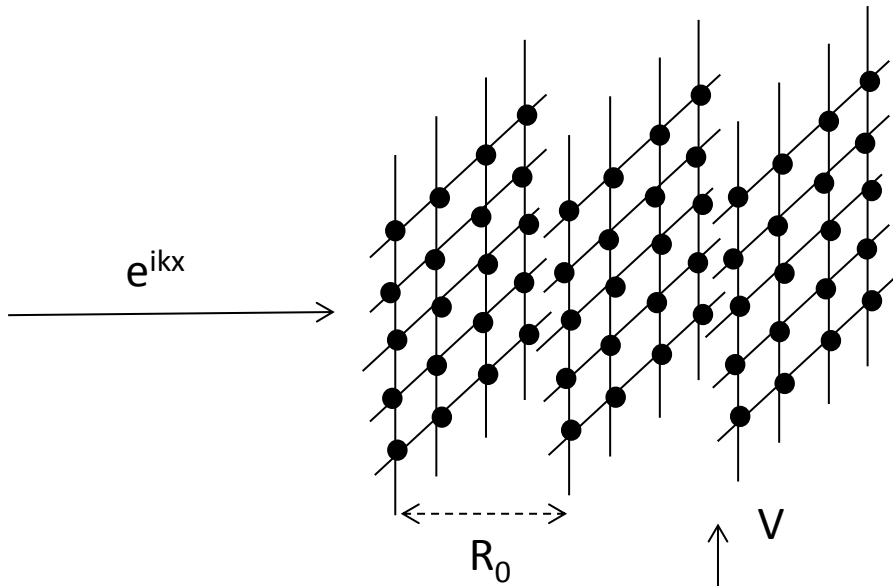
$$n=1 \text{ št. stanj } 6 \cdot 2 n^2 = 12$$

$$n=2 \text{ št. stanj } 6 \cdot 2 n^2 = 48$$



vodikov „kristal“:
 10^{22} atomov / cm^3

e^- v periodičnem potencialu



močno sipanje za $k_n = n\pi/R_0$

$$e^{ik_n x} \rightarrow \cos(k_n x), \sin(k_n x)$$

$$W_+ = \frac{\hbar^2 k_n^2}{2m_e} + A^2 \int V(x) \cos^2(k_n x) dx$$

$$W_- = \frac{\hbar^2 k_n^2}{2m_e} + A^2 \int V(x) \sin^2(k_n x) dx$$

