

Scintillation Detectors

Particle Detection via Luminescence



Scintillators – General Characteristics

Principle:

dE/dx converted into visible light

Detection via photosensor

[e.g. photomultiplier, human eye ...]

Main Features:

Sensitivity to energy

Fast time response

Pulse shape discrimination

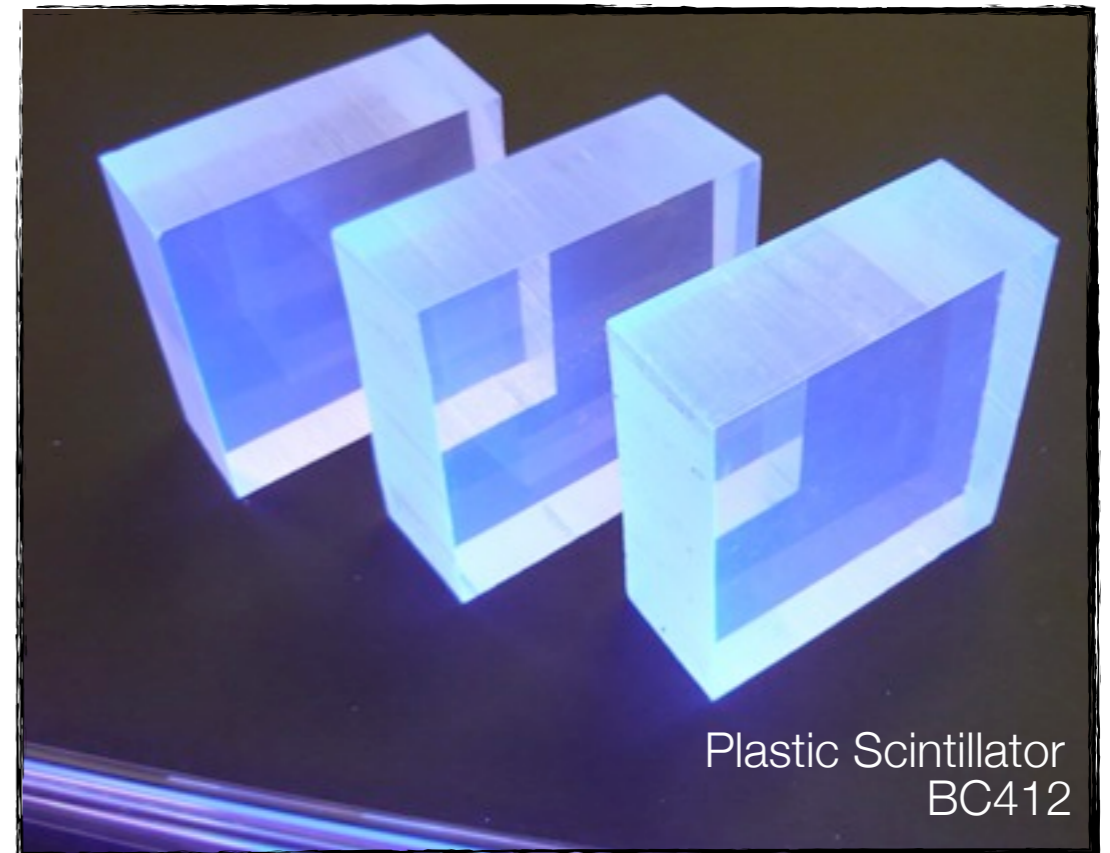
Requirements

High efficiency for conversion of excitation energy to fluorescent radiation

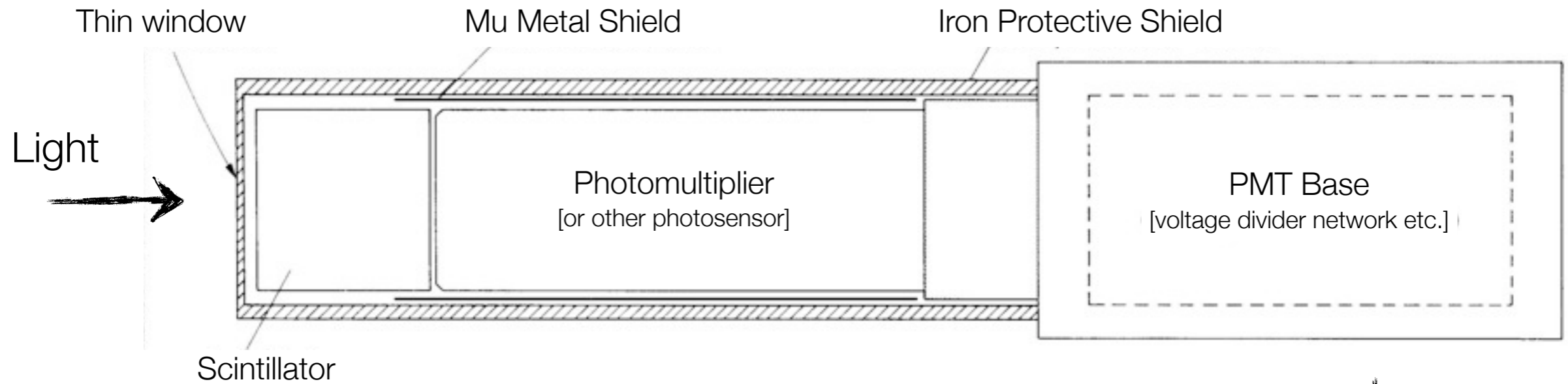
Transparency to its fluorescent radiation to allow transmission of light

Emission of light in a spectral range detectable for photosensors

Short decay time to allow fast response



Scintillators – Basic Counter Setup

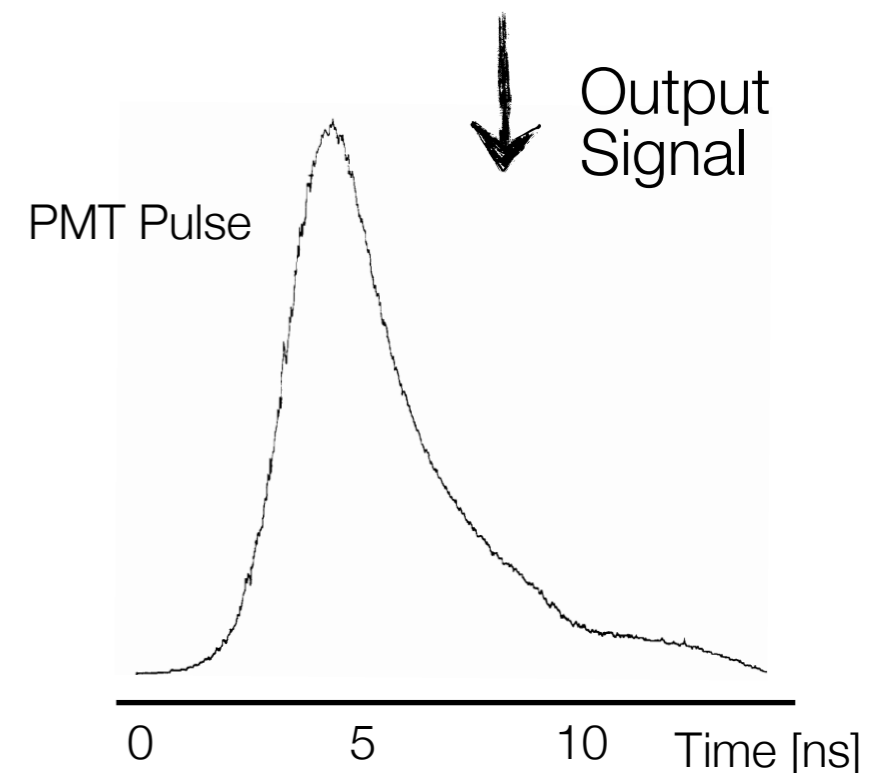


Scintillator Types:

Photosensors

- Photomultipliers
- Micro-Channel Plates
- Hybrid Photo Diodes
- Visible Light Photon Counter
- Silicon Photomultipliers

- Organic Scintillators
- Inorganic Crystals
- Gases



Inorganic Crystals

Materials:

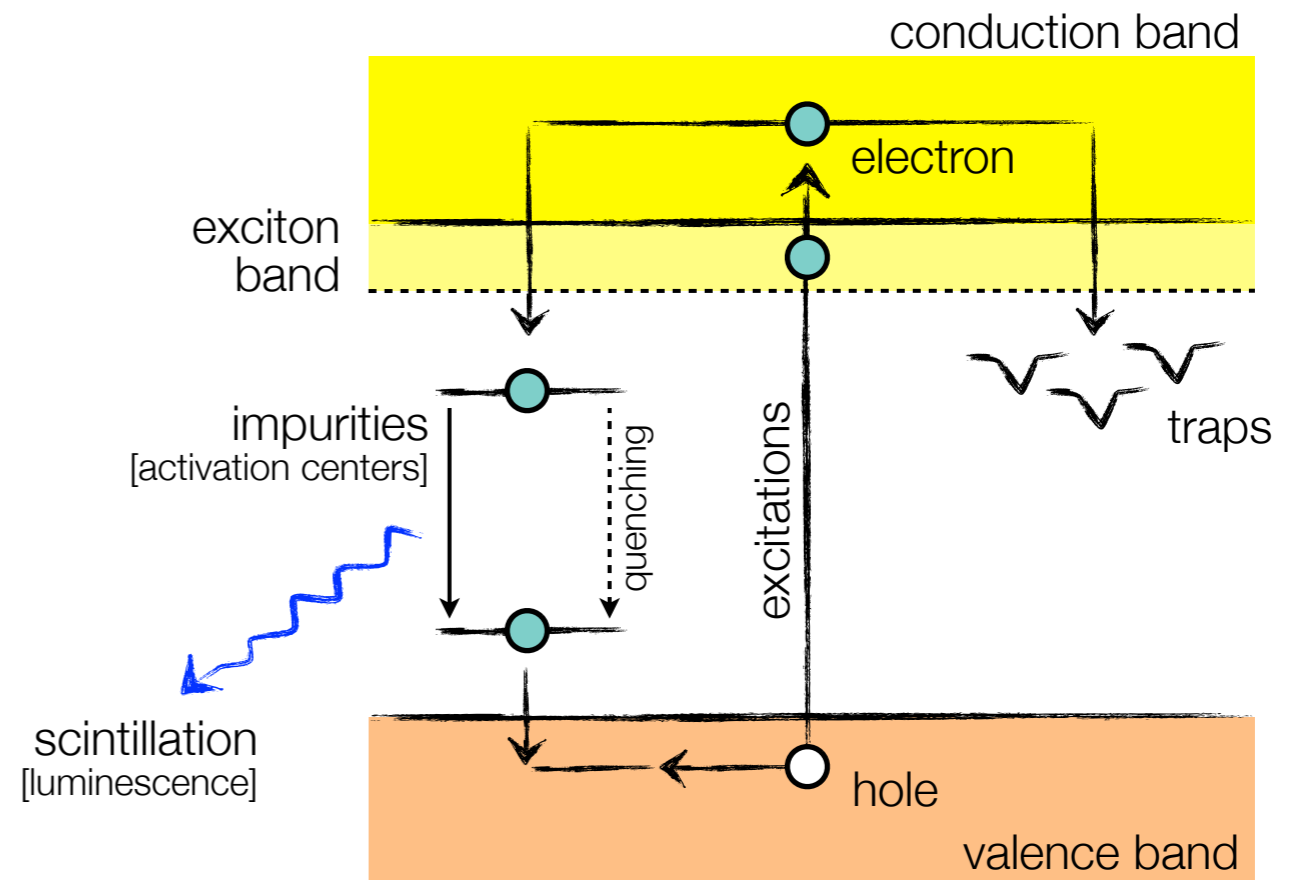
Sodium iodide (NaI)
Cesium iodide (CsI)
Barium fluoride (BaF₂)
...

Mechanism:

Energy deposition by ionization
Energy transfer to impurities
Radiation of scintillation photons

Time constants:

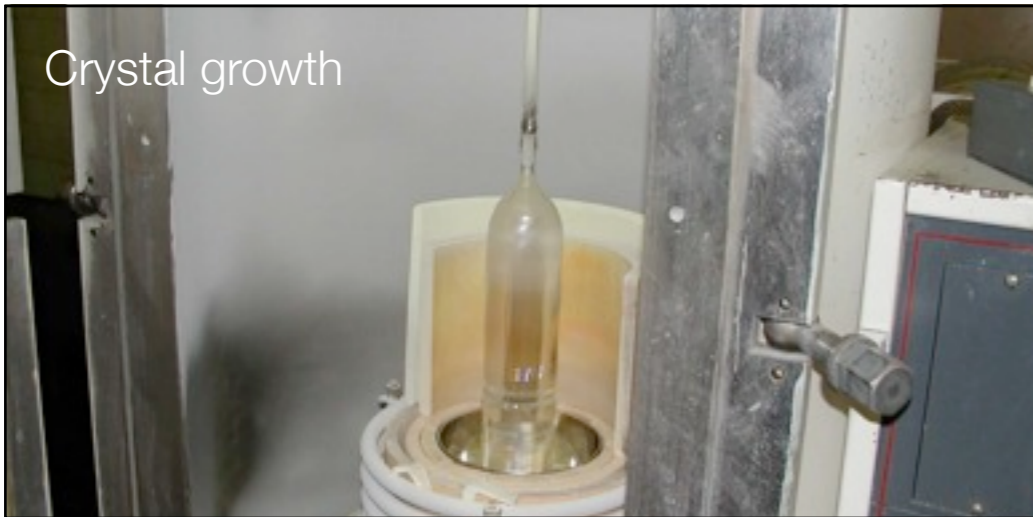
Fast: recombination from activation centers [ns ... μs]
Slow: recombination due to trapping [ms ... s]



Energy bands in
impurity activated crystal
showing excitation, luminescence,
quenching and trapping

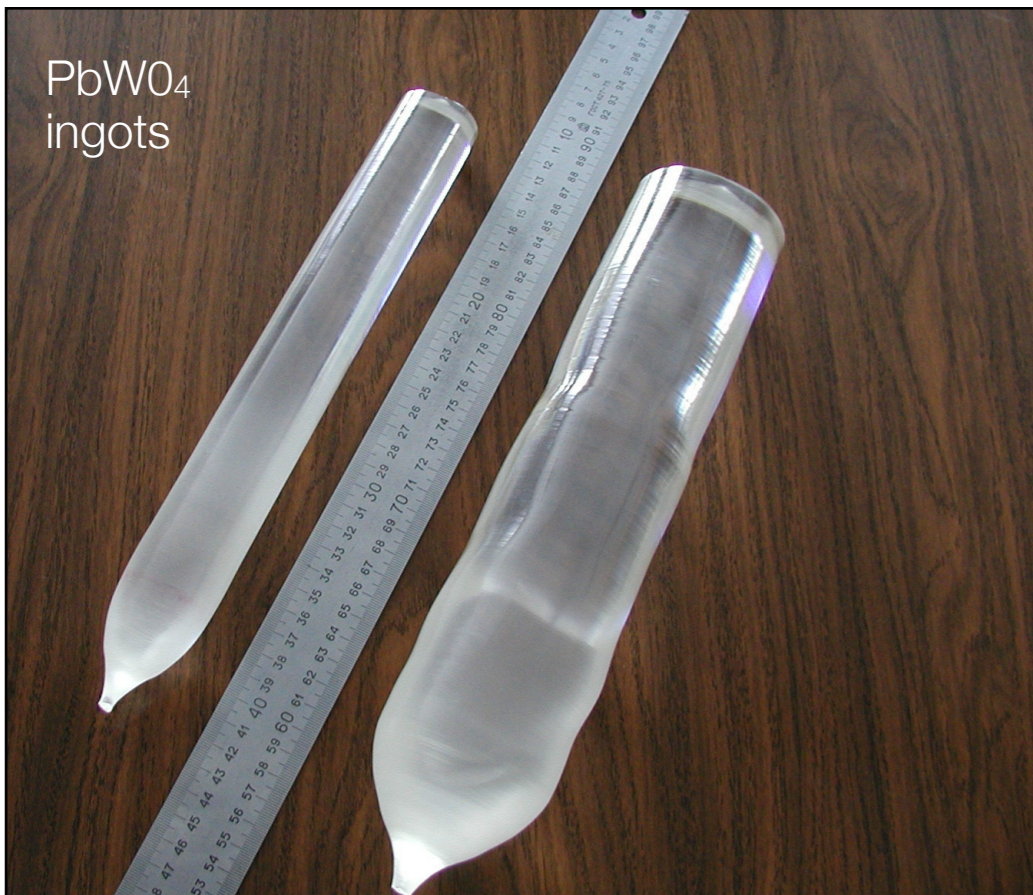
Inorganic Crystals

Crystal growth

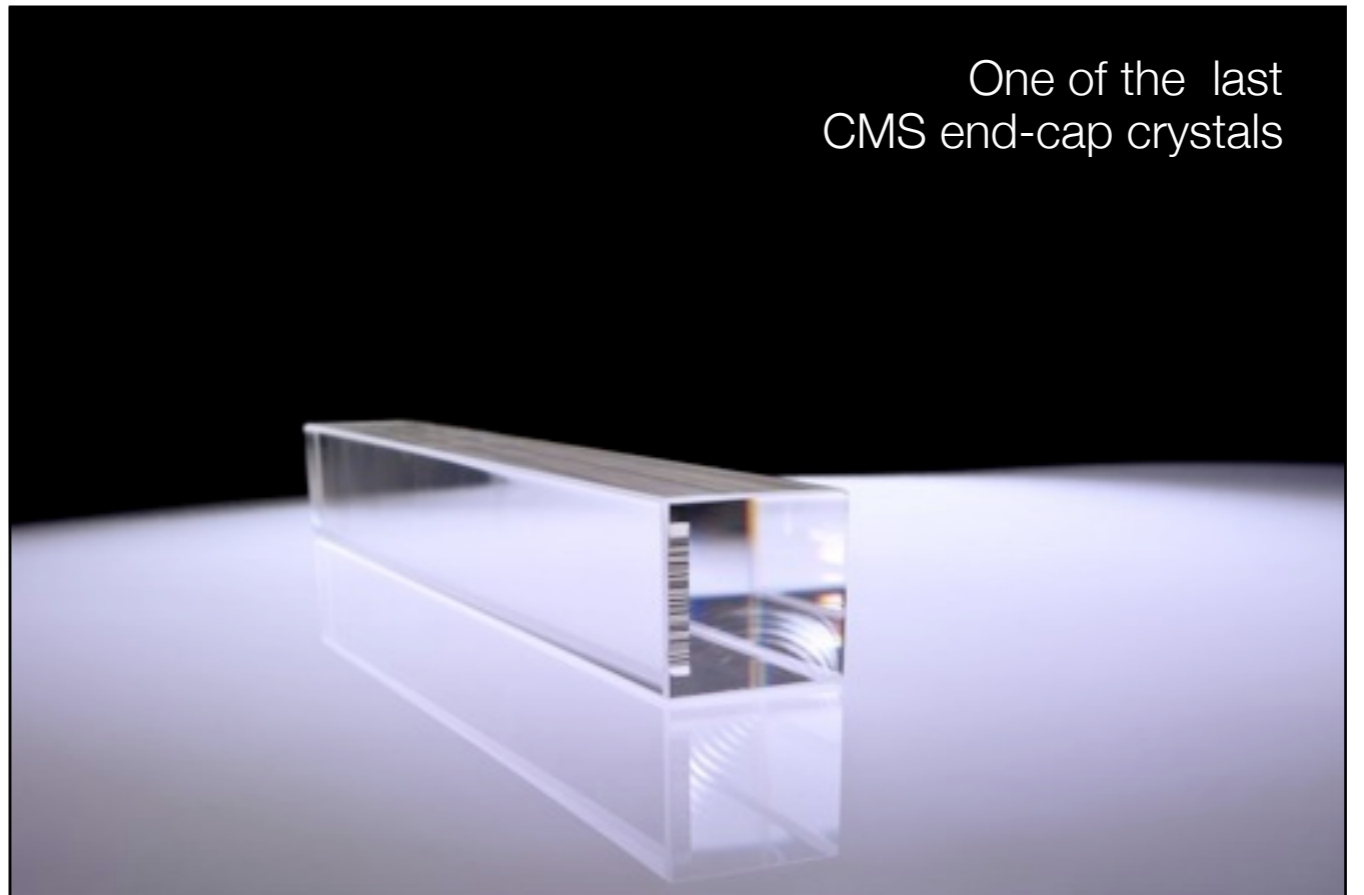


Example CMS
Electromagnetic Calorimeter

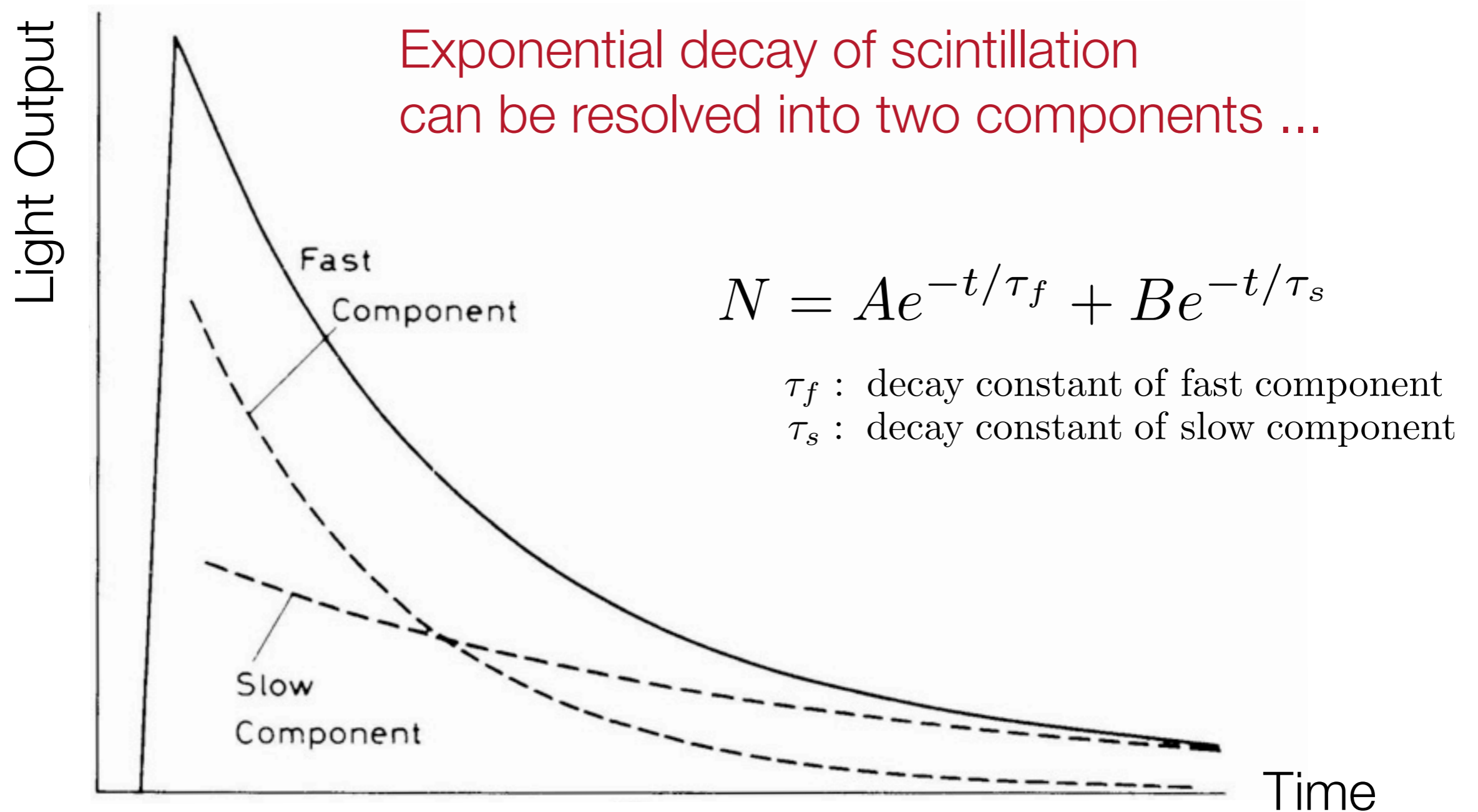
PbWO₄
ingots



One of the last
CMS end-cap crystals

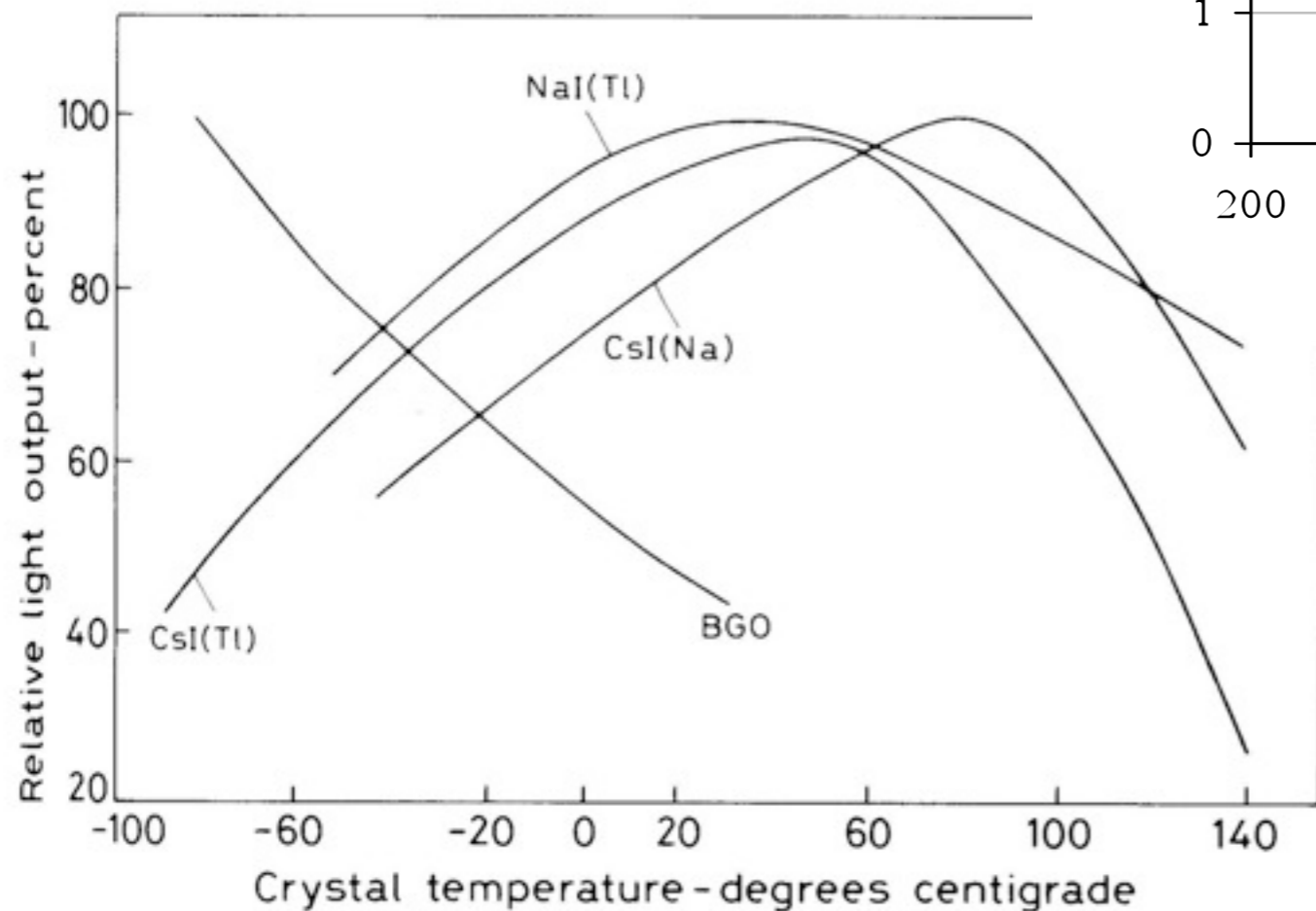
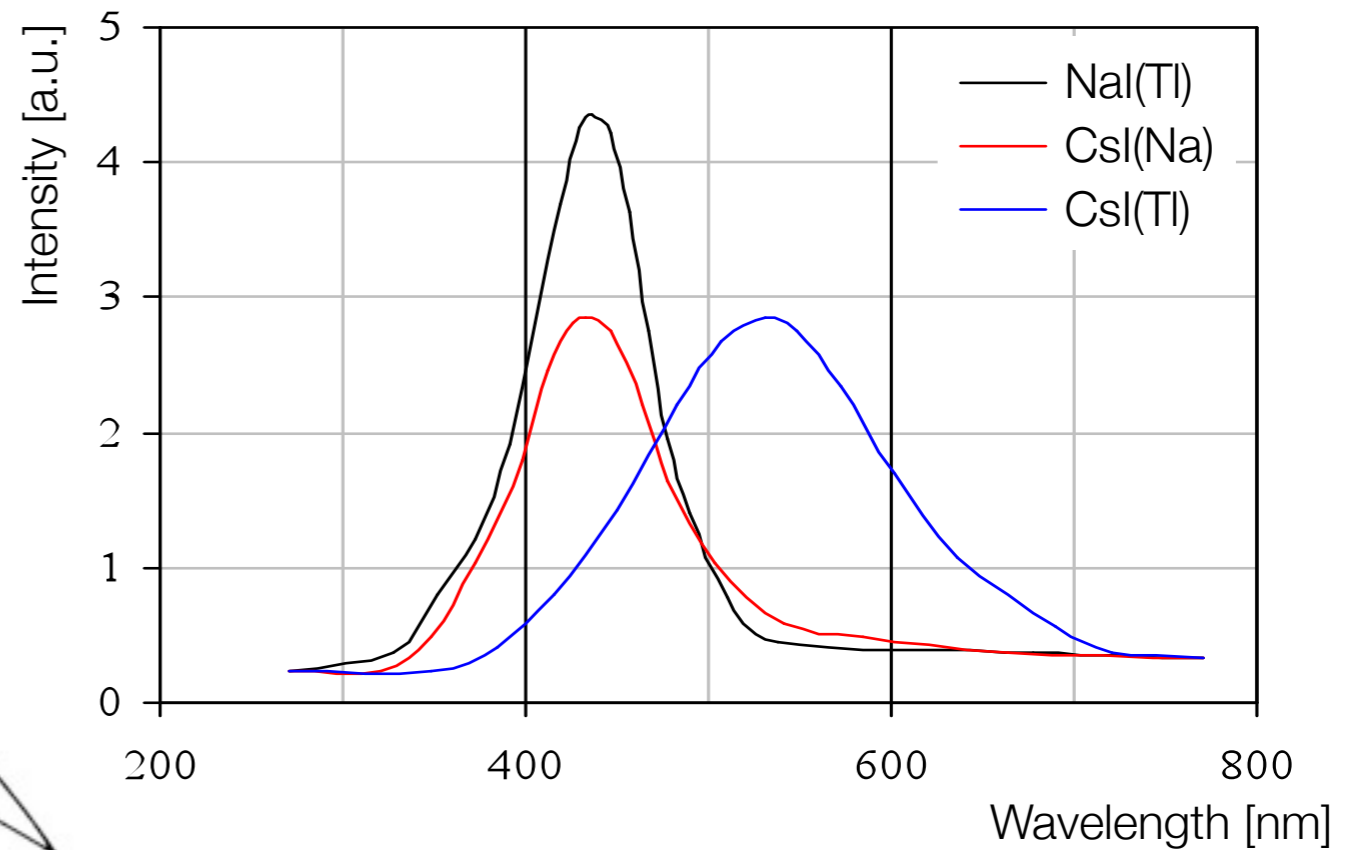


Inorganic Crystals – Time Constants



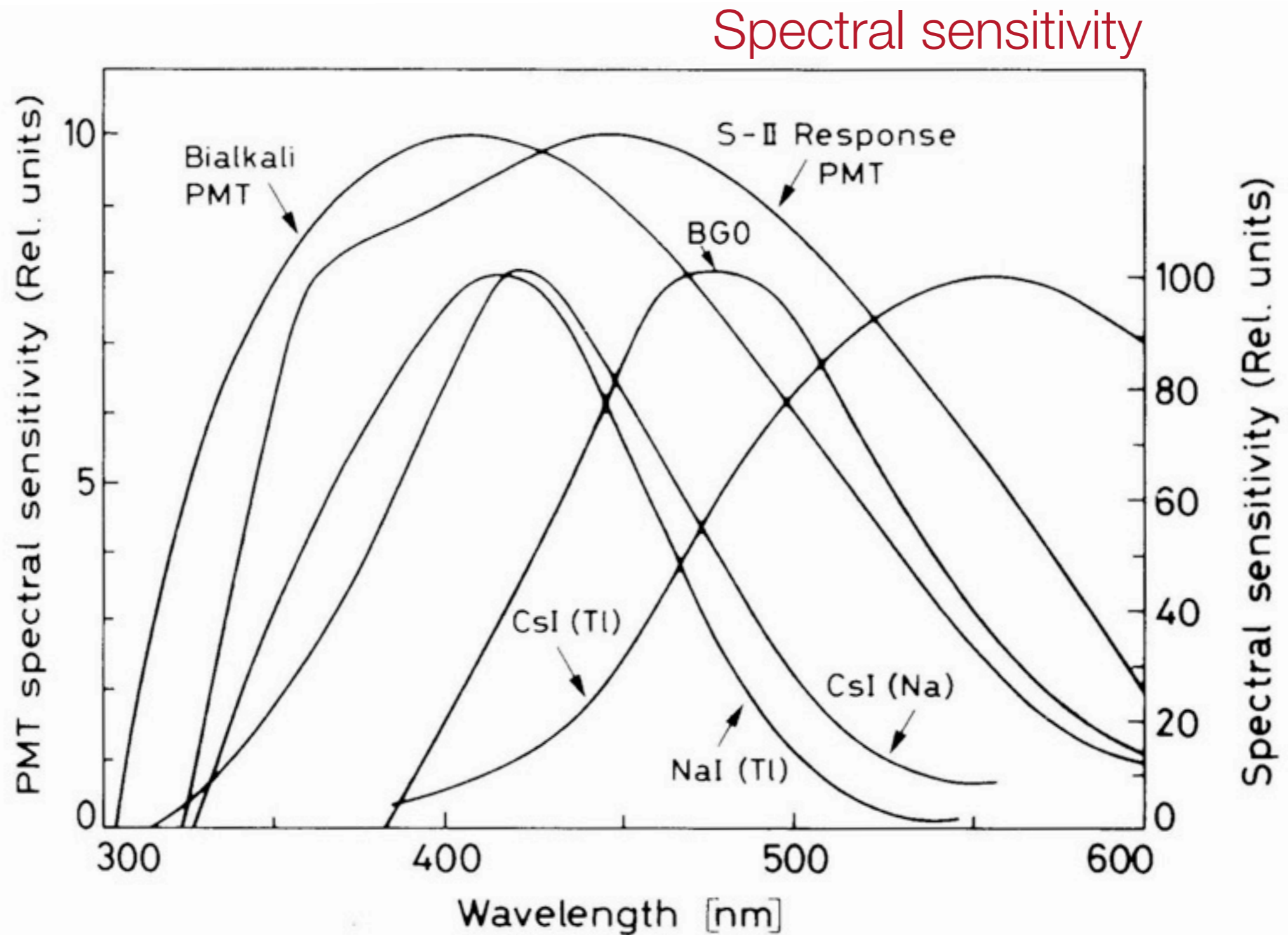
Inorganic Crystals – Light Output

Scintillation Spectrum for NaI and CsI



**Strong
Temperature Dependence**
[in contrast to organic scintillators]

Inorganic Crystals – Light Output & PMT Sensitivity



Scintillation in Liquid Nobel Gases

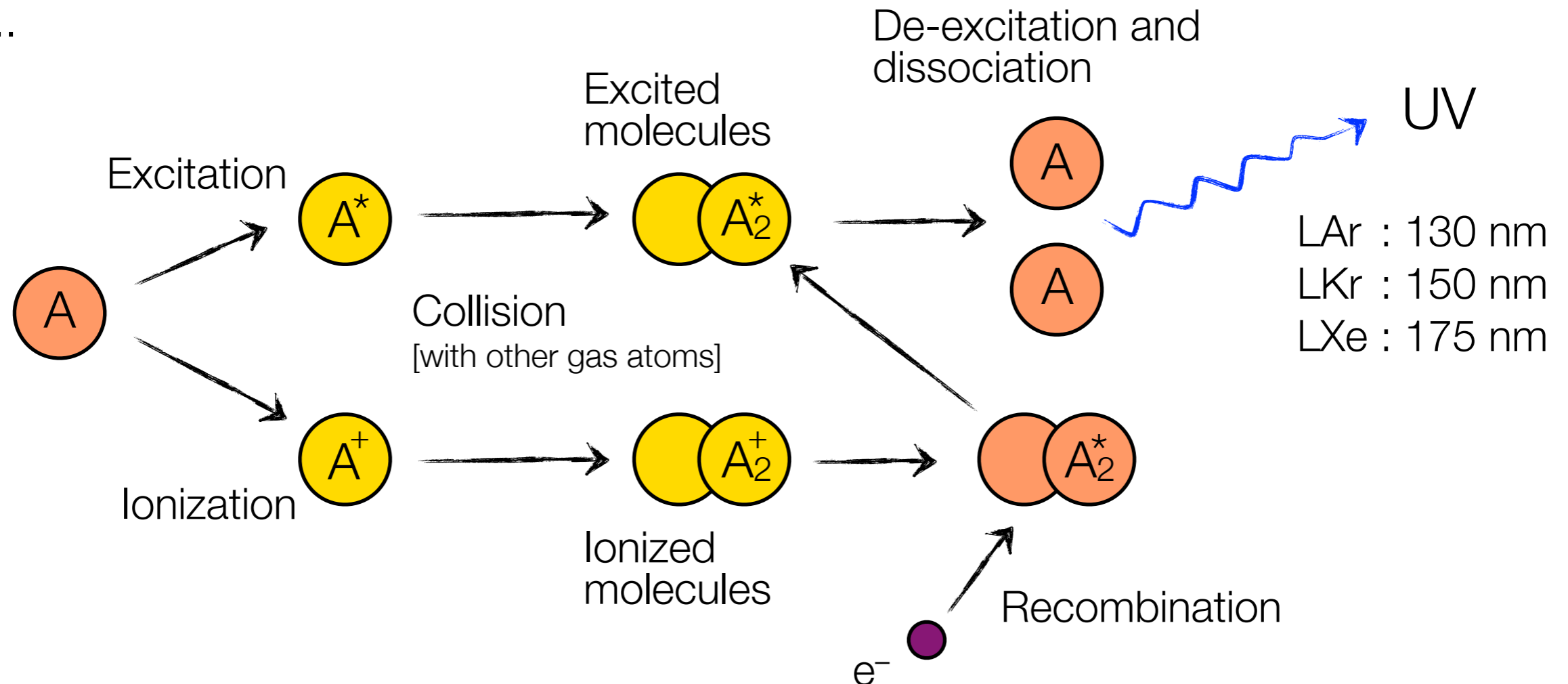
Materials:

Helium (He)
Liquid Argon (LAr)
Liquid Xenon (LXe)
...

Decay time constants:

Helium : $\tau_1 = .02 \mu\text{s}$, $\tau_2 = 3 \mu\text{s}$

Argon : $\tau_1 \leq .02 \mu\text{s}$



Inorganic Scintillators – Properties

Scintillator material	Density [g/cm ³]	Refractive Index	Wavelength [nm] for max. emission	Decay time constant [μs]	Photons/MeV
NaI	3.7	1.78	303	0.06	$8 \cdot 10^4$
NaI(Tl)	3.7	1.85	410	0.25	$4 \cdot 10^4$
CsI(Tl)	4.5	1.80	565	1.0	$1.1 \cdot 10^4$
Bi ₄ Ge ₃ O ₁₂	7.1	2.15	480	0.30	$2.8 \cdot 10^3$
CsF	4.1	1.48	390	0.003	$2 \cdot 10^3$
LSO	7.4	1.82	420	0.04	$1.4 \cdot 10^4$
PbWO ₄	8.3	1.82	420	0.006	$2 \cdot 10^2$
LHe	0.1	1.02	390	0.01/1.6	$2 \cdot 10^2$
LAr	1.4	1.29*	150	0.005/0.86	$4 \cdot 10^4$
LXe	3.1	1.60*	150	0.003/0.02	$4 \cdot 10^4$

* at 170 nm

Inorganic Scintillators – Properties

Numerical examples:

NaI(Tl)

$$\begin{aligned}\lambda_{\max} &= 410 \text{ nm}; h\nu = 3 \text{ eV} \\ \text{photons/MeV} &= 40000 \\ \tau &= 250 \text{ ns}\end{aligned}$$

PBWO₄

$$\begin{aligned}\lambda_{\max} &= 420 \text{ nm}; h\nu = 3 \text{ eV} \\ \text{photons/MeV} &= 200 \\ \tau &= 6 \text{ ns}\end{aligned}$$

Scintillator quality:

Light yield – ϵ_{sc} \equiv fraction of energy loss going into photons

e.g. NaI(Tl) : 40000 photons; 3 eV/photon $\rightarrow \epsilon_{\text{sc}} = 4 \cdot 10^4 \cdot 3 \text{ eV} / 10^6 \text{ eV} = 11.3\%$
PBWO₄ : 200 photons; 3 eV/photon $\rightarrow \epsilon_{\text{sc}} = 2 \cdot 10^2 \cdot 3 \text{ eV} / 10^6 \text{ eV} = 0.06\%$

[for 1 MeV particle]

Organic Scintillators

Aromatic hydrocarbon compounds:

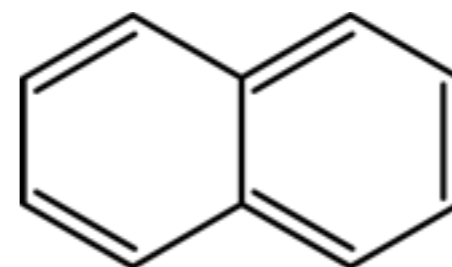
e.g. Naphthalene [C₁₀H₈]
Anthracene [C₁₄H₁₀]
Stilbene [C₁₄H₁₂]
...

Very fast!
[Decay times of O(ns)]

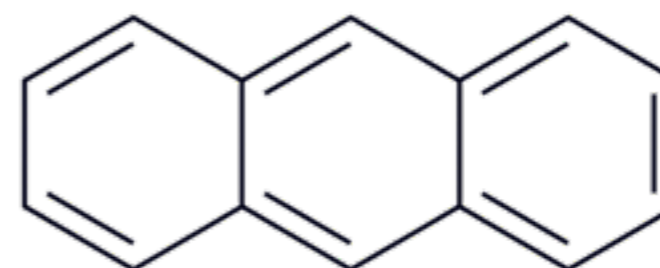
Scintillation light arises from delocalized electrons in π -orbitals ...

Transitions of 'free' electrons ...

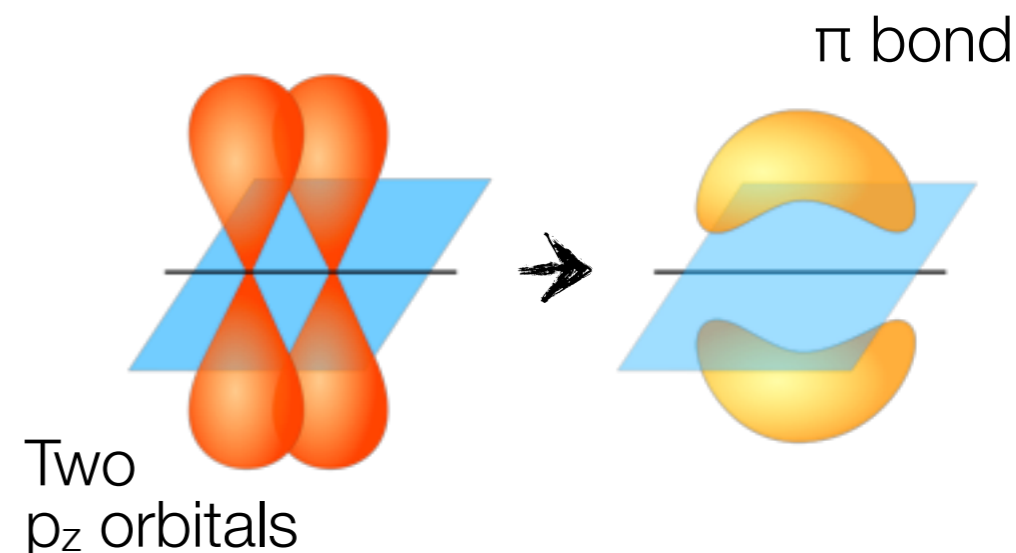
Naphthalene



Anthracene



Scintillation is based on electrons of the C=C bond ...



Organic Scintillators

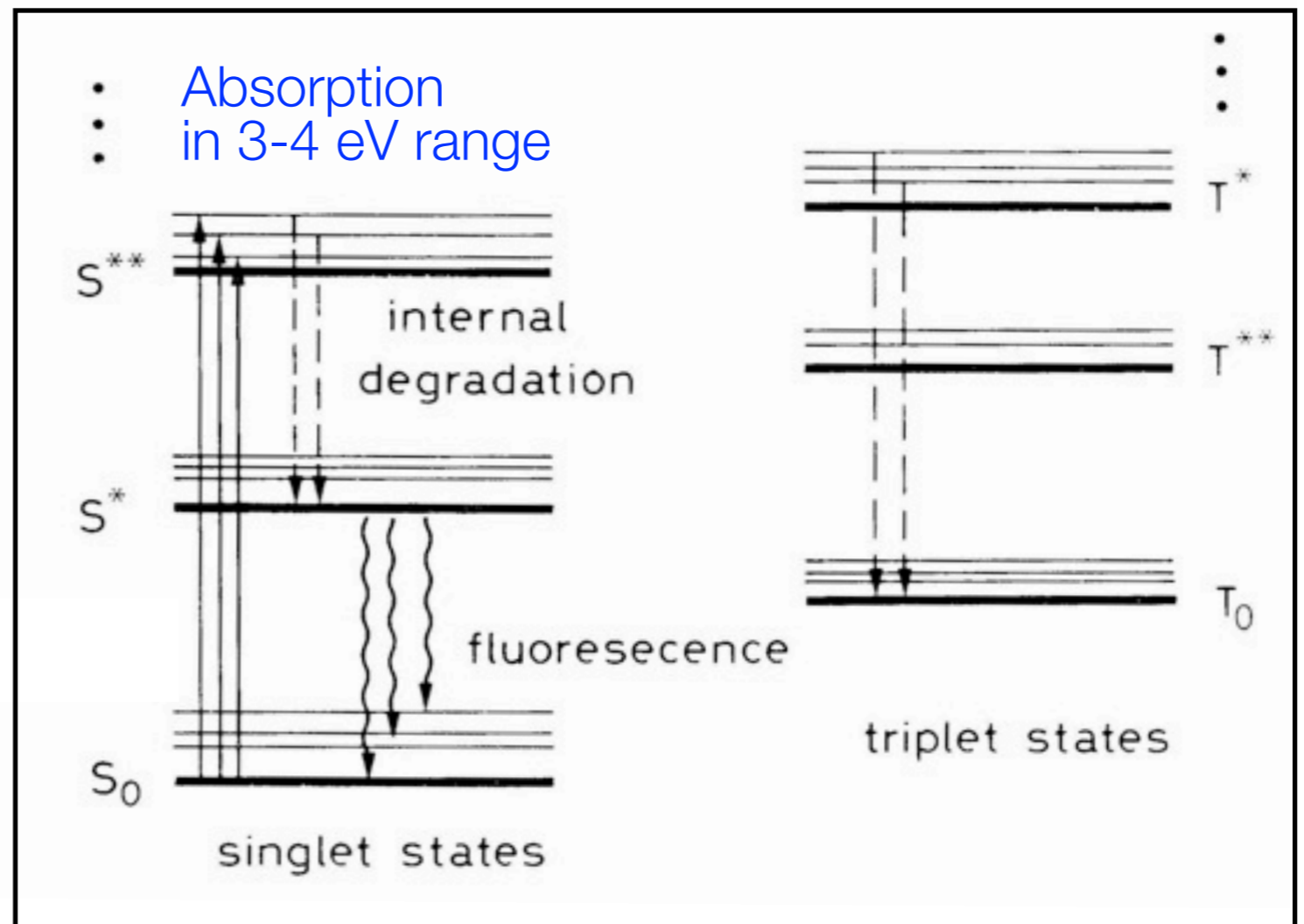
Molecular states:

Singlet states

Triplet states

Fluorescence in
UV range
[~ 320 nm]

➔ usage of
wavelength shifters



Fluorescence : $S_1 \rightarrow S_0$ [$< 10^{-8}$ s]

Phosphorescence : $T_0 \rightarrow S_0$ [$> 10^{-4}$ s]

Organic Scintillators

Transparency requires:

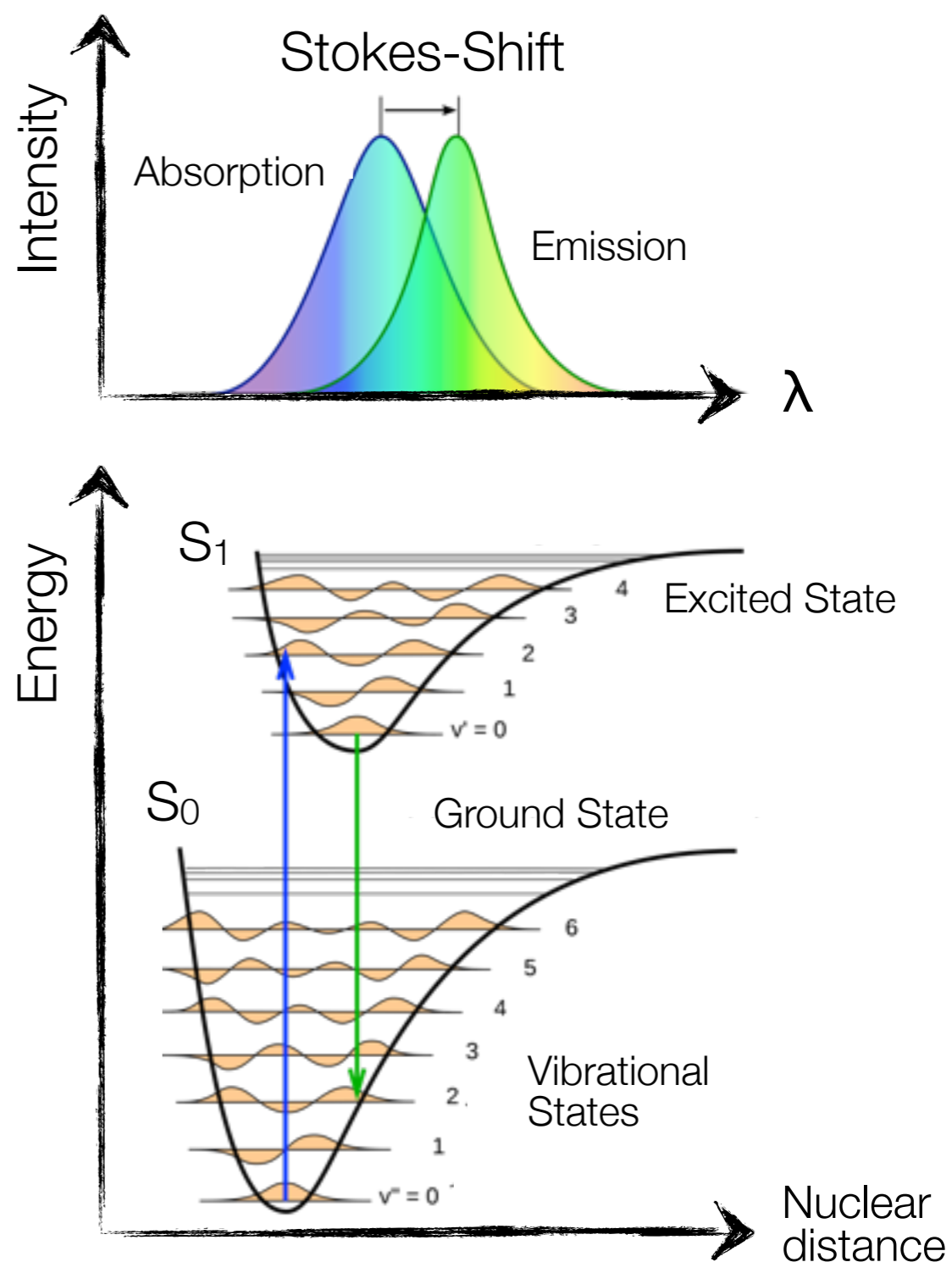
Shift of absorption
and emission spectra ...

Shift due to

Franck-Condon Principle

Excitation into higher vibrational states
De-excitation from lowest vibrational state

Excitation time scale : 10^{-14} s
Vibrational time scale : 10^{-12} s
 S_1 lifetime : 10^{-8} s



Plastic and Liquid Scintillators

In practice use ...

solution of organic scintillators
[solved in plastic or liquid]

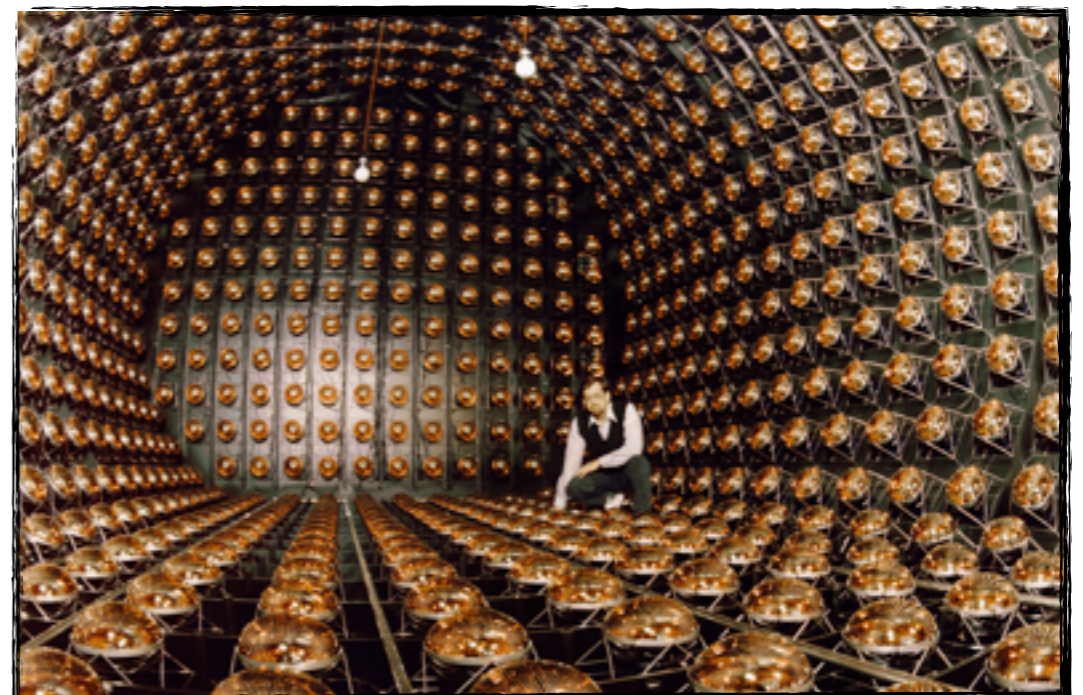
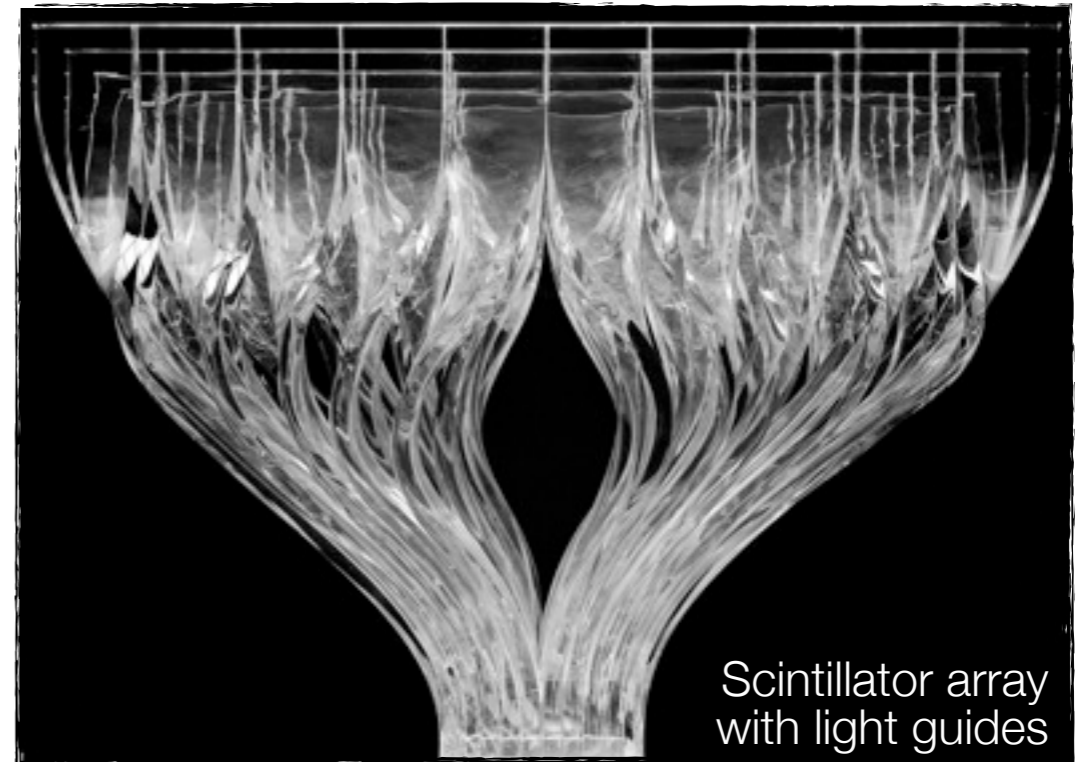
- + large concentration of primary 'fluor'
- + smaller concentration of secondary 'fluor'
- + ...

Scintillator requirements:

Solvable in base material

High fluorescence yield

Absorption spectrum must overlap
with emission spectrum of base material

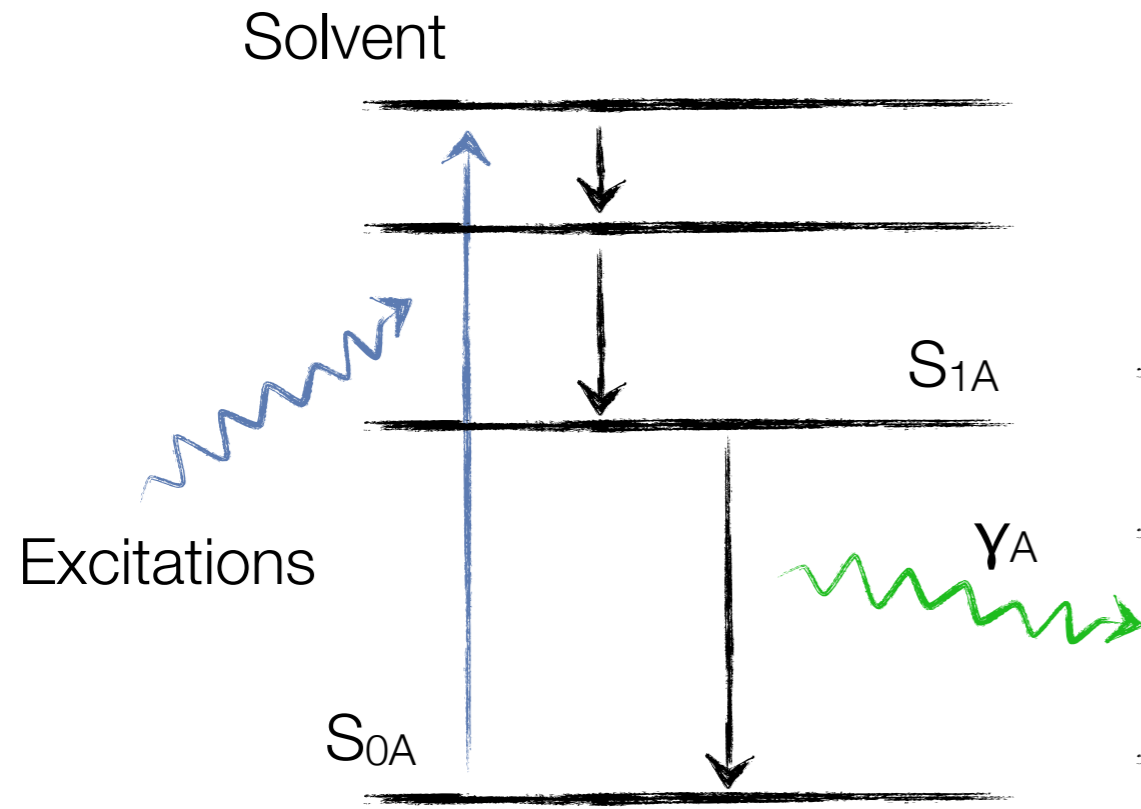


LSND experiment

Plastic and Liquid Scintillators

A

Energy deposit in base material \rightarrow excitation



Primary fluorescent

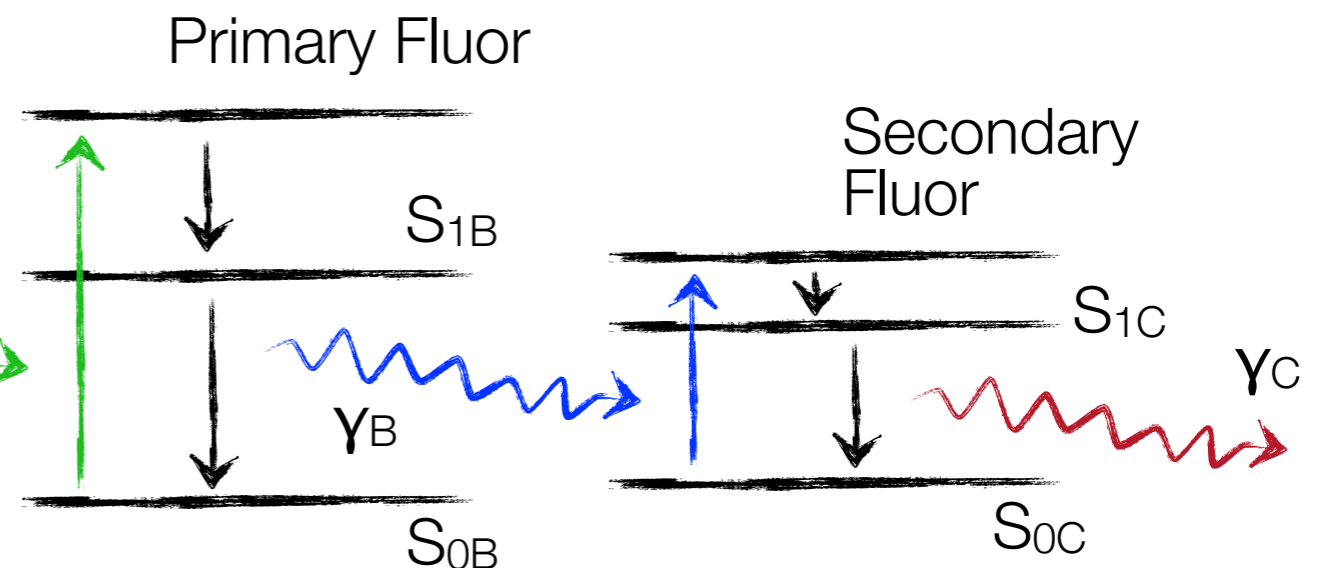
- Good light yield ...
- Absorption spectrum matched to excited states in base material ...

B

Secondary fluorescent

C

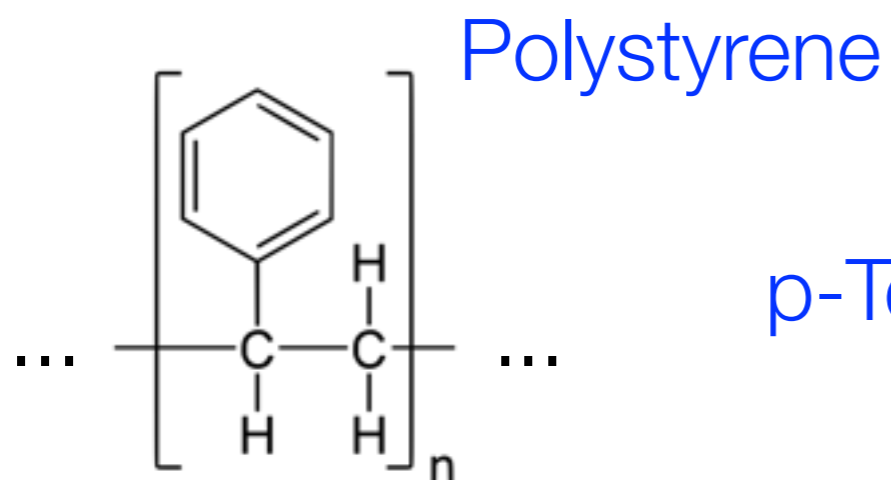
Wave length shifter



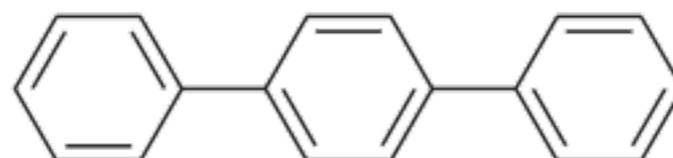
Plastic and Liquid Scintillators

Some widely used solvents and solutes

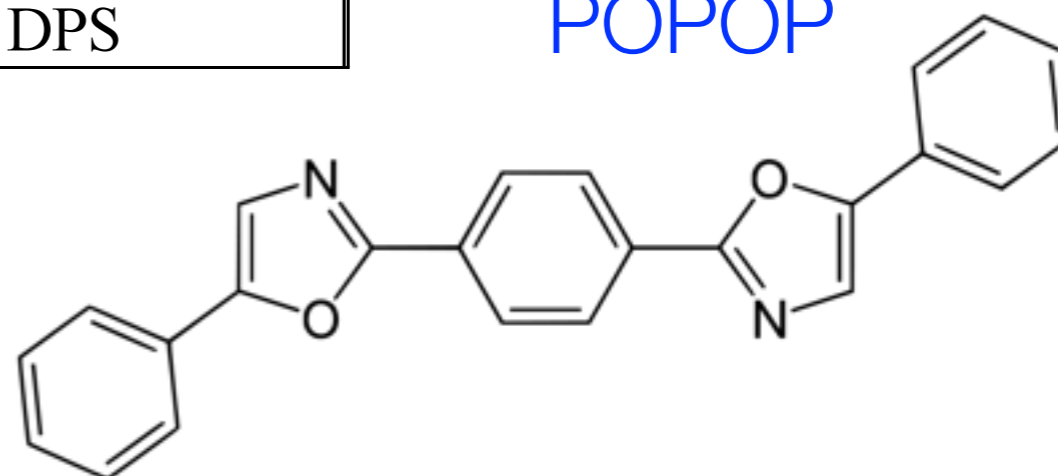
	solvent	secondary fluor	tertiary fluor
Liquid scintillators	Benzene Toluene Xylene	p-terphenyl DPO PBD	POPOP BBO BPO
Plastic scintillators	Polyvinylbenzene Polyvinyltoluene Polystyrene	p-terphenyl DPO PBD	POPOP TBP BBO DPS



p-Terphenyl



POPOP



Wavelength Shifting

Principle:

Absorption of primary scintillation light

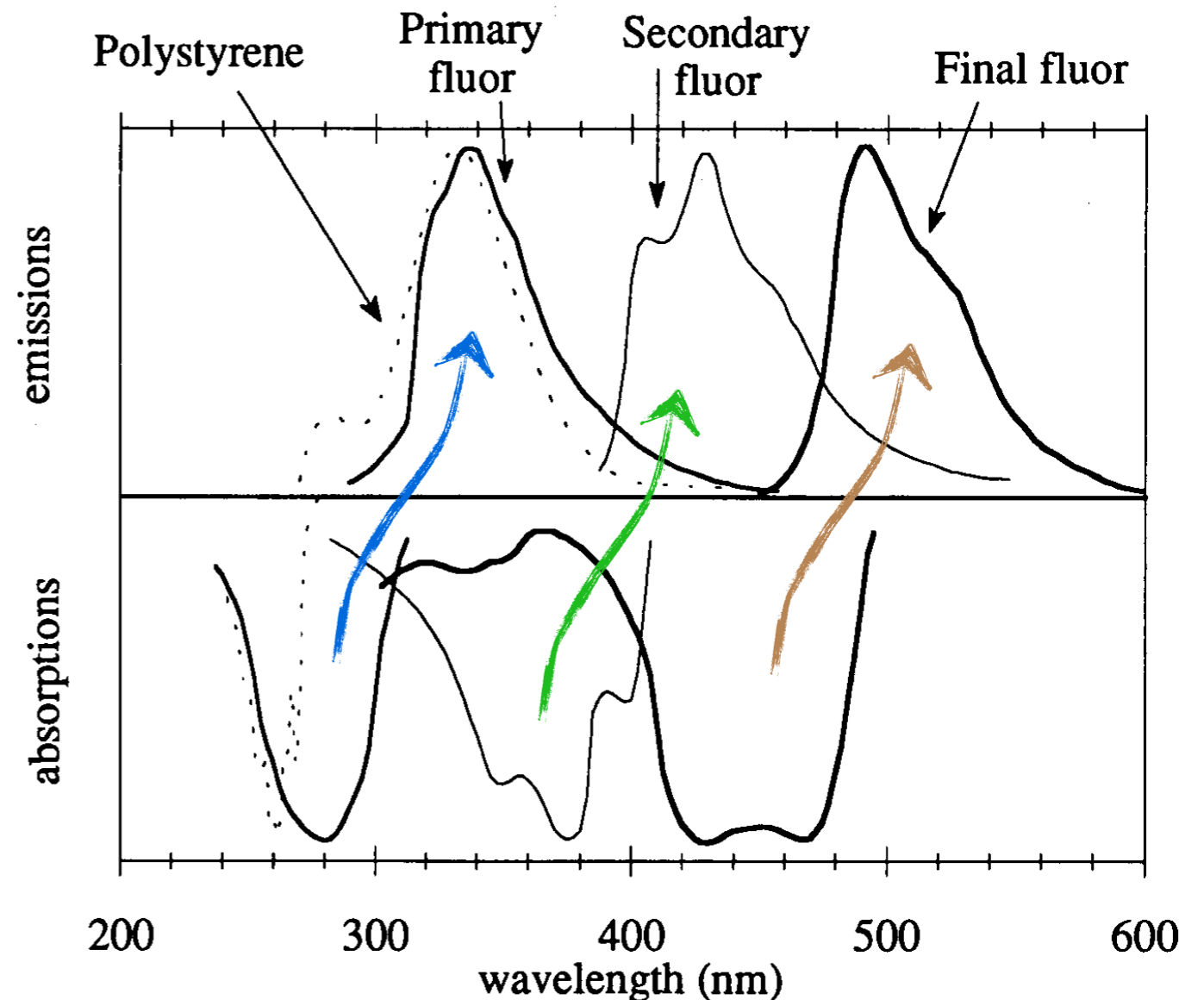
Re-emission at longer wavelength

Adapts light to spectral sensitivity of photosensor

Requirement:

Good transparency for emitted light

Schematics of wavelength shifting principle

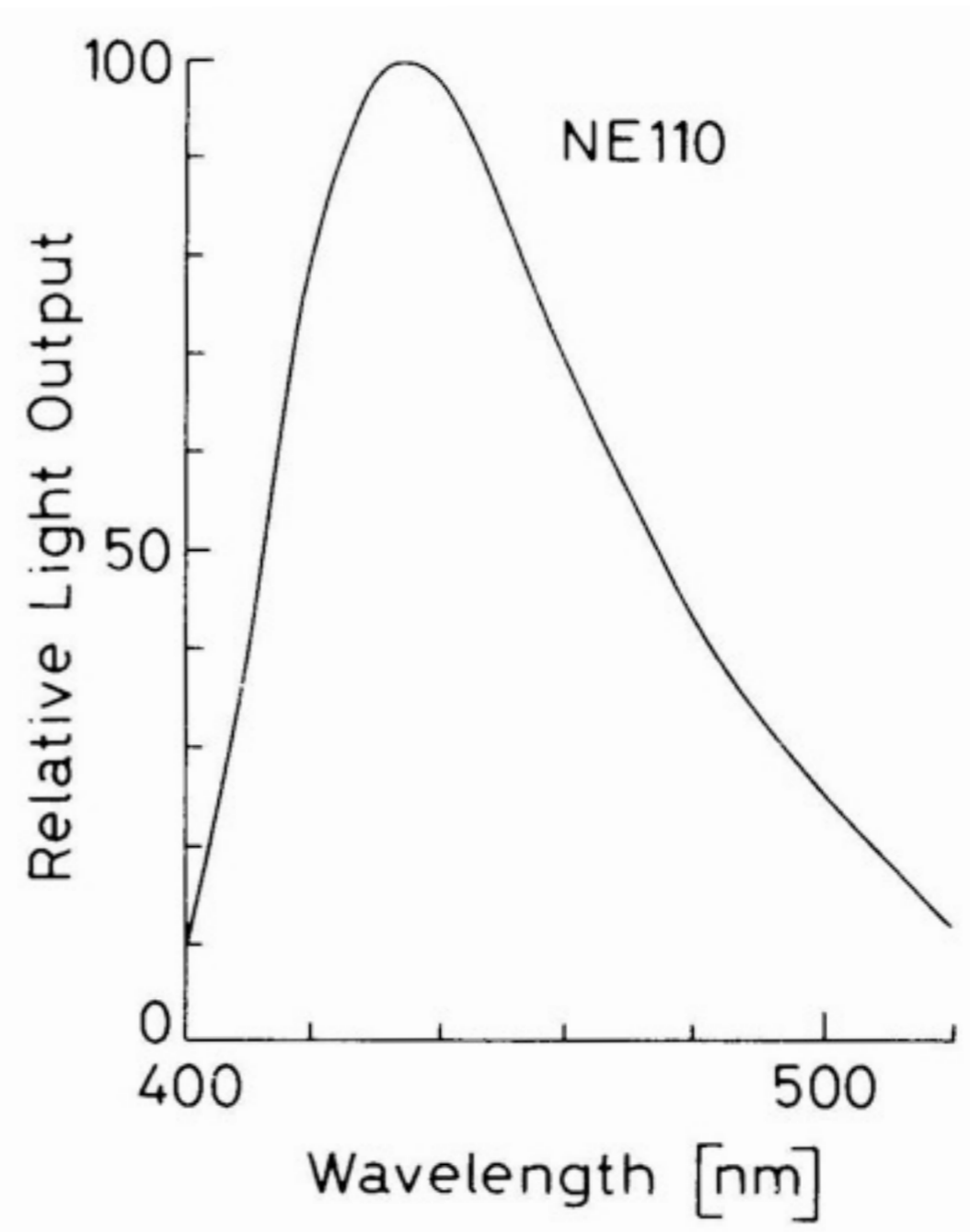
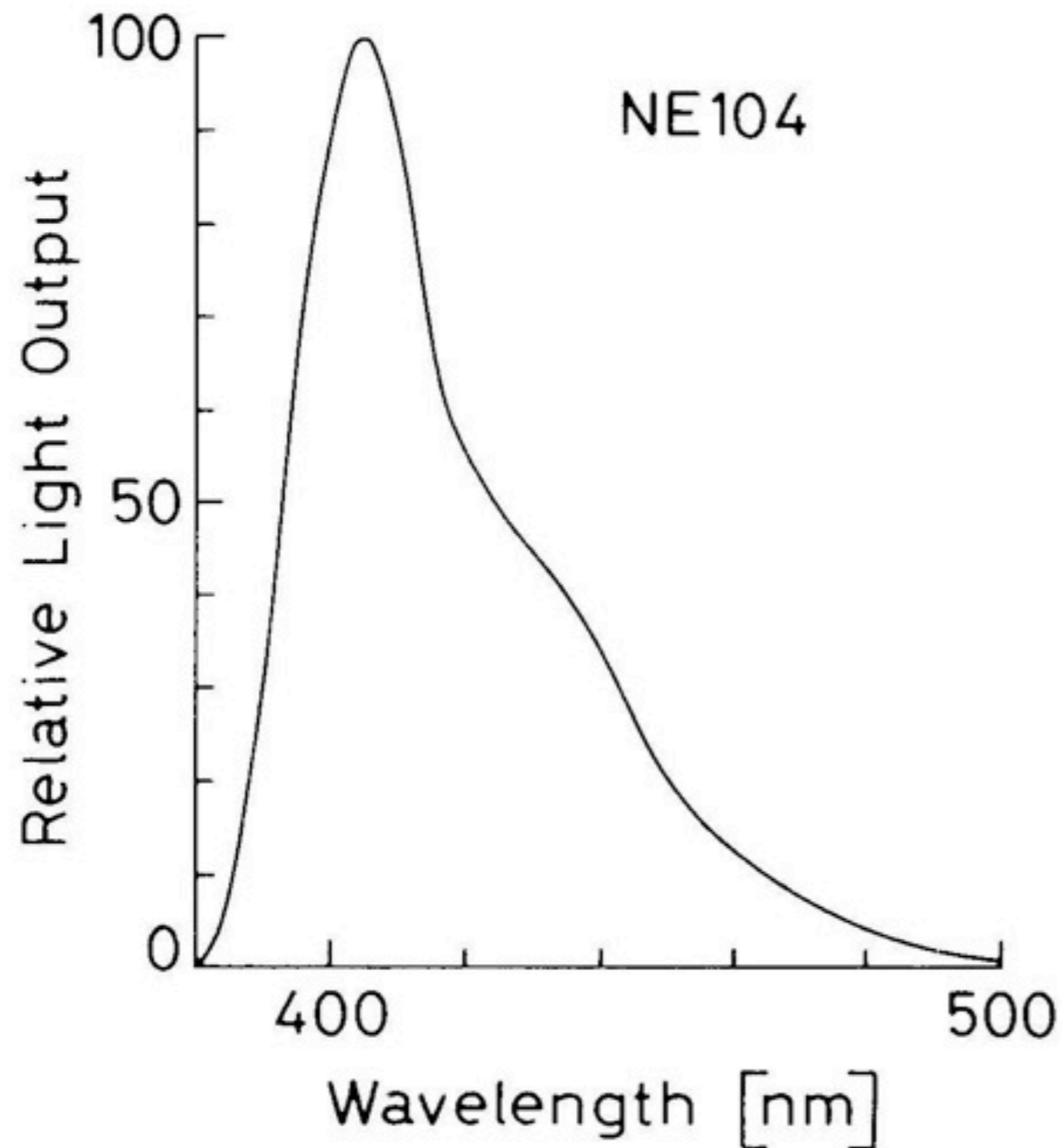


Organic Scintillators – Properties

Scintillator material	Density [g/cm ³]	Refractive Index	Wavelength [nm] for max. emission	Decay time constant [ns]	Photons/MeV
Naphtalene	1.15	1.58	348	11	$4 \cdot 10^3$
Antracene	1.25	1.59	448	30	$4 \cdot 10^4$
p-Terphenyl	1.23	1.65	391	6-12	$1.2 \cdot 10^4$
NE102*	1.03	1.58	425	2.5	$2.5 \cdot 10^4$
NE104*	1.03	1.58	405	1.8	$2.4 \cdot 10^4$
NE110*	1.03	1.58	437	3.3	$2.4 \cdot 10^4$
NE111*	1.03	1.58	370	1.7	$2.3 \cdot 10^4$
BC400**	1.03	1.58	423	2.4	$2.5 \cdot 10^2$
BC428**	1.03	1.58	480	12.5	$2.2 \cdot 10^4$
BC443**	1.05	1.58	425	2.2	$2.4 \cdot 10^4$

* Nuclear Enterprises, U.K.
 ** Bicron Corporation, USA

Organic Scintillators – Properties



Organic Scintillators – Properties

Light yield:
[without quenching]

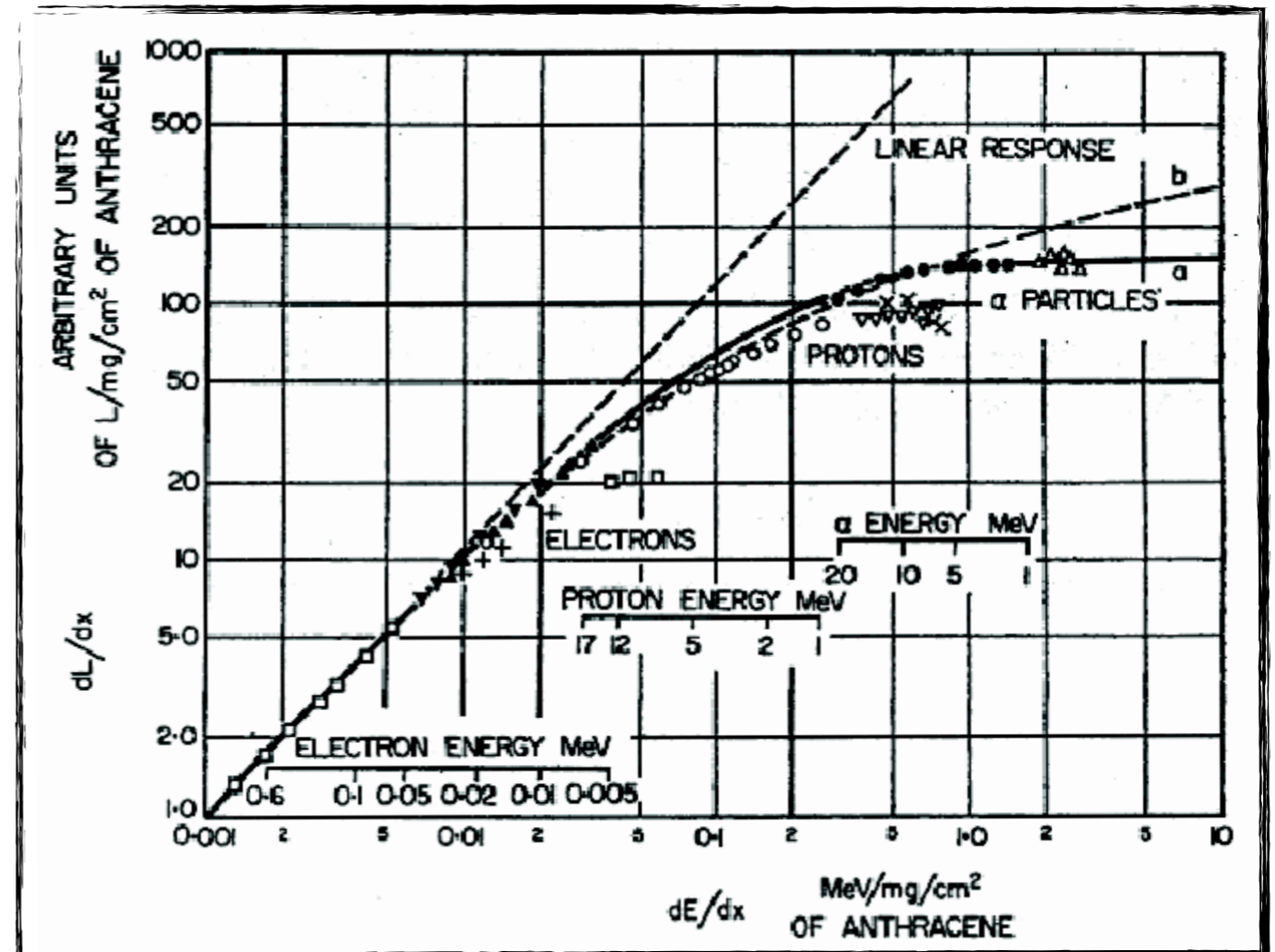
$$\frac{dL}{dx} = L_0 \frac{dE}{dx}$$

Quenching:
non-linear response due to
saturation of available states

Birk's law:

$$\frac{dL}{dx} = L_0 \frac{\frac{dE}{dx}}{1 + kB \frac{dE}{dx}}$$

[kB needs to be determined experimentally]



Also other
parameterizations ...

Response different
for different particle types ...

Scintillators – Comparison

Inorganic Scintillators

Advantages

high light yield [typical; $\epsilon_{sc} \approx 0.13$]
high density [e.g. $PbWO_4$: 8.3 g/cm³]
good energy resolution

Disadvantages

complicated crystal growth
large temperature dependence

Expensive

Organic Scintillators

Advantages

very fast
easily shaped
small temperature dependence
pulse shape discrimination possible

Disadvantages

lower light yield [typical; $\epsilon_{sc} \approx 0.03$]
radiation damage

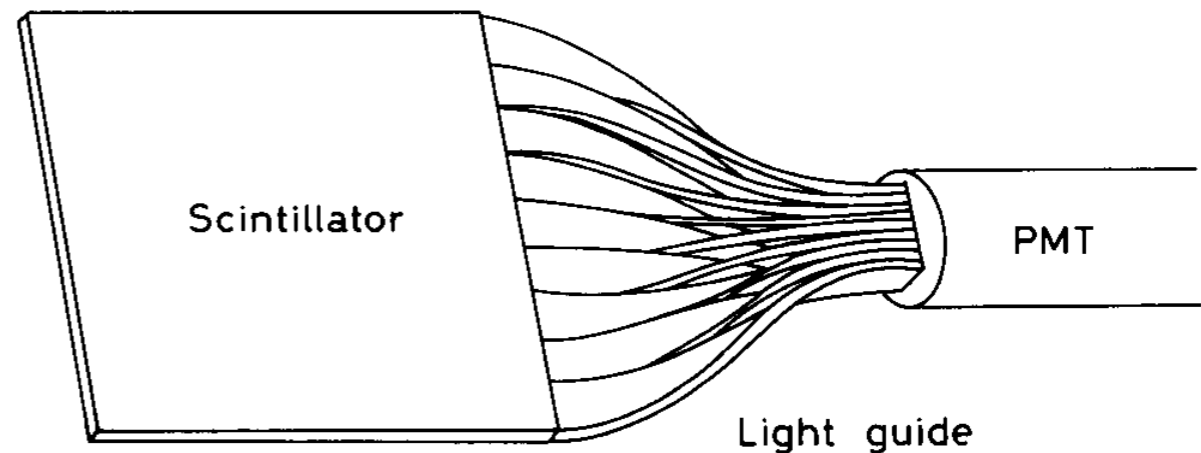
Cheap

Scintillation Counters – Setup

Scintillator light to be guided to photosensor

- Light guide
[Plexiglas; optical fibers]

Light transfer by
total internal reflection
[maybe combined with wavelength shifting]



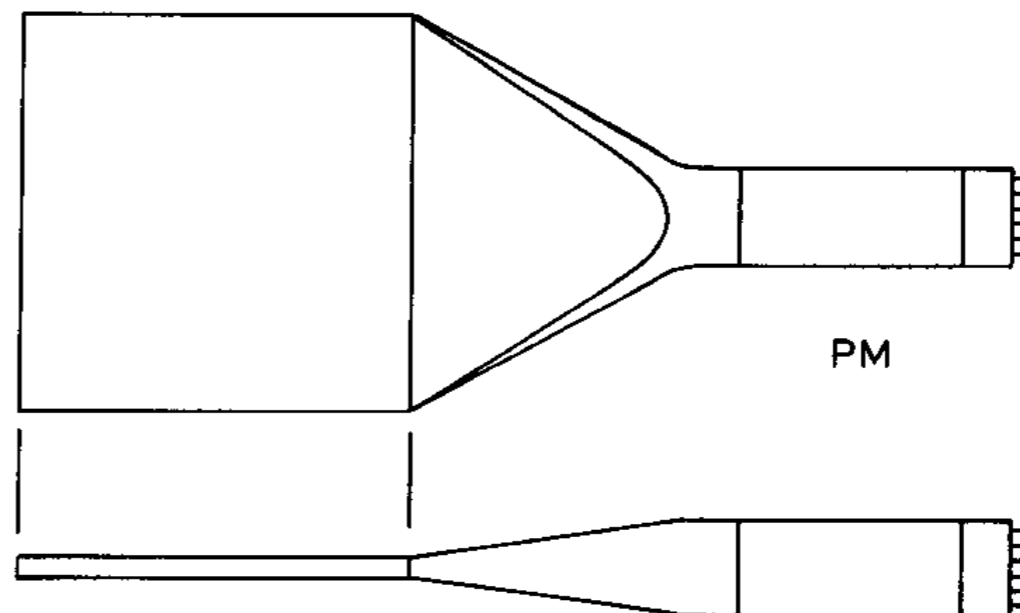
Light guide

Liouville's Theorem:

Complete light transfer
impossible as $\Delta x \Delta \theta = \text{const.}$
[limits acceptance angle]

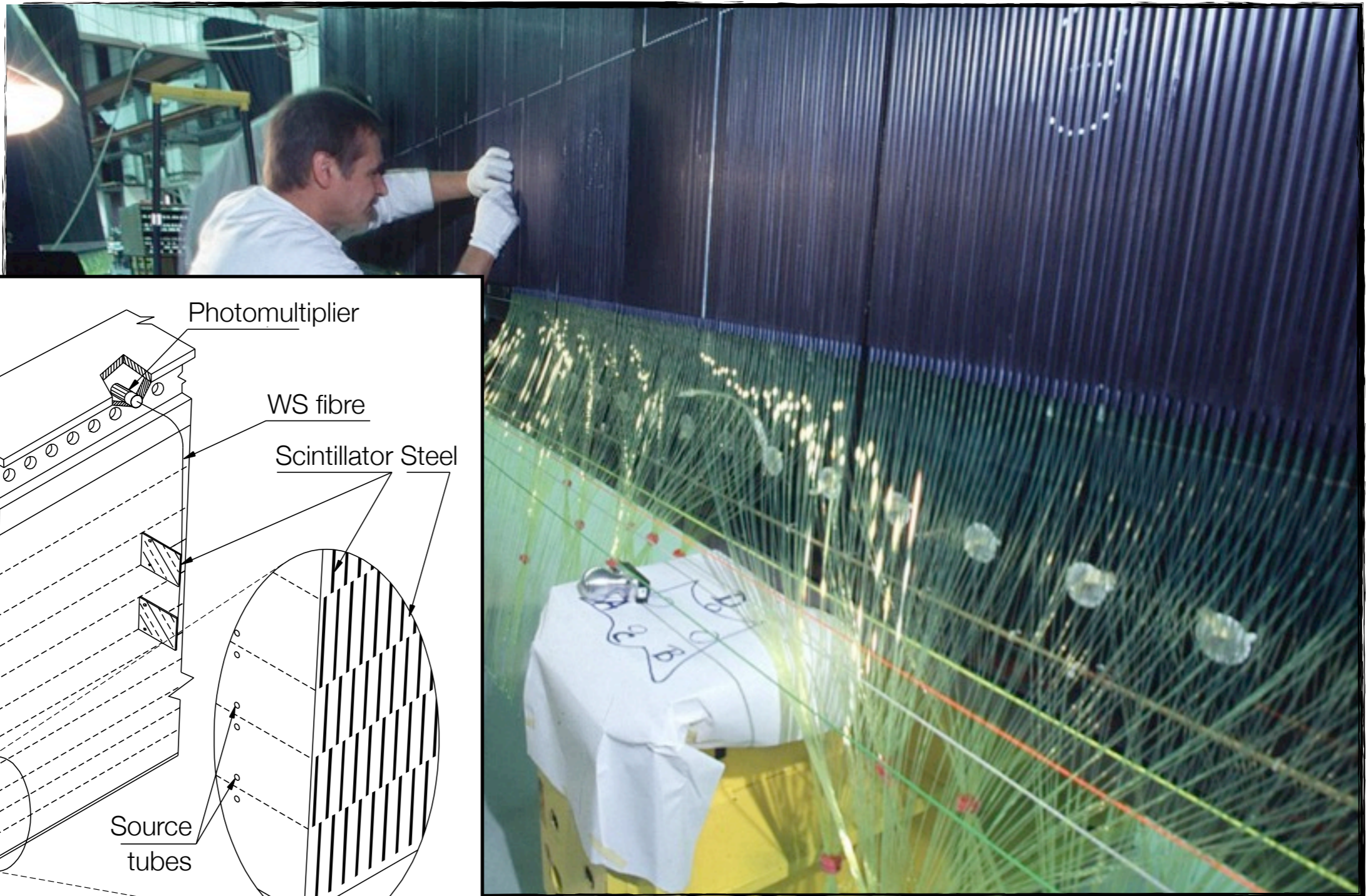
Use adiabatic light guide
like 'fish tail';

- appreciable energy loss

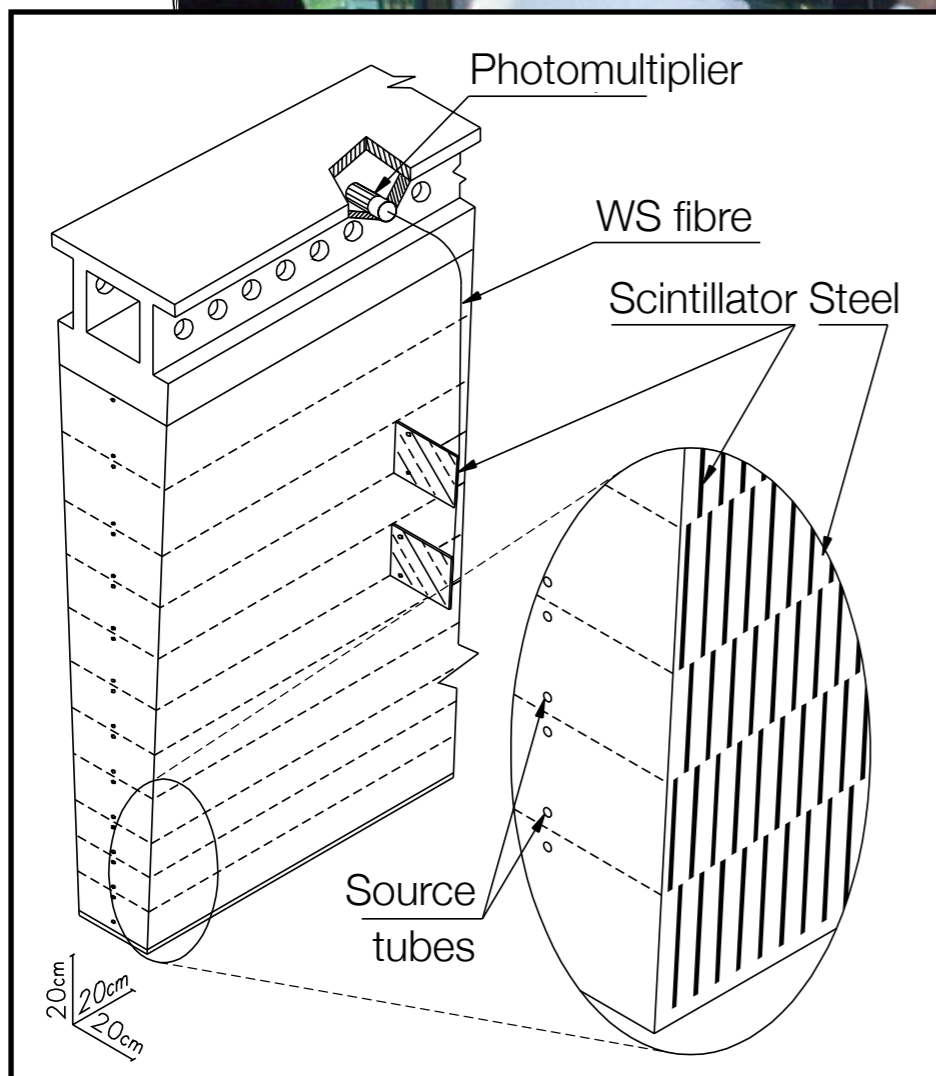


'fish tail'

Scintillation Counters – Setup



ATLAS Tile Calorimeter



Photon Detection

Purpose : Convert light into a detectable electronic signal

Principle : Use **photo-electric effect** to convert photons to **photo-electrons (p.e.)**

Requirement :

High **Photon Detection Efficiency (PDE)** or **Quantum Efficiency**; $Q.E. = N_{p.e.}/N_{photons}$

Available devices [Examples]:

Photomultipliers [PMT]

Micro Channel Plates [MCP]

Photo Diodes [PD]

Hybrid Photo Diodes [HPD]

Visible Light Photon Counters [VLPC]

Silicon Photomultipliers [SiPM]

Photomultipliers

Principle:

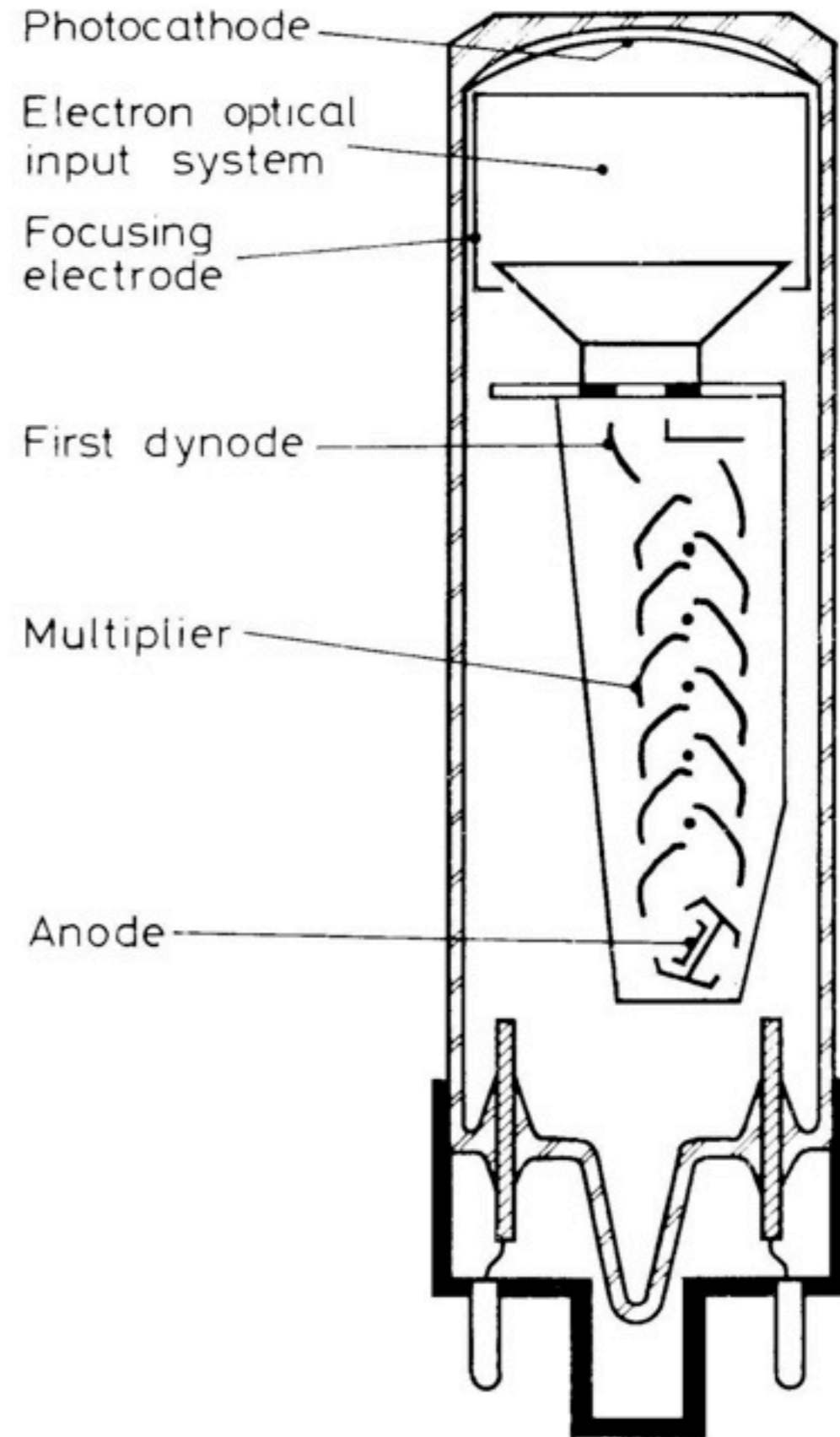
Electron emission
from photo cathode

Secondary emission
from dynodes; dynode gain: 3-50 [f(E)]

Typical PMT Gain: $> 10^6$
[PMT can see single photons ...]



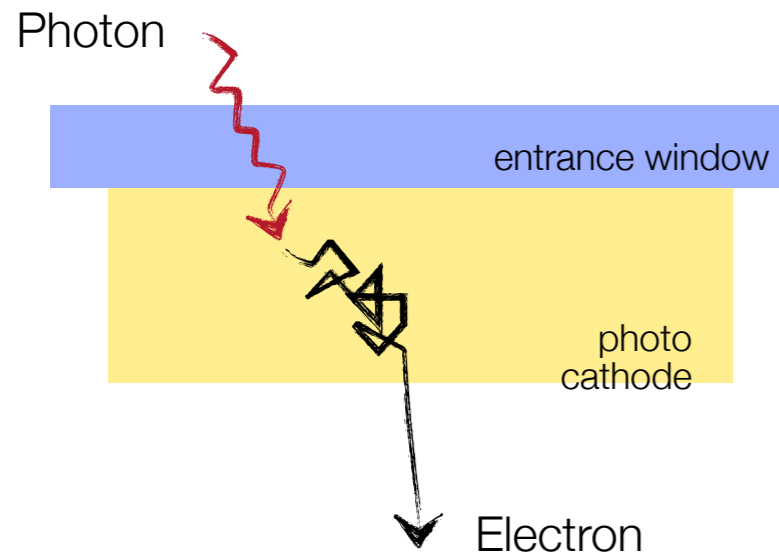
PMT
Collection



Photomultipliers – Photocathode

Bialkali: SbRbCs ; SbK_2Cs

γ -conversion
via photo effect ...

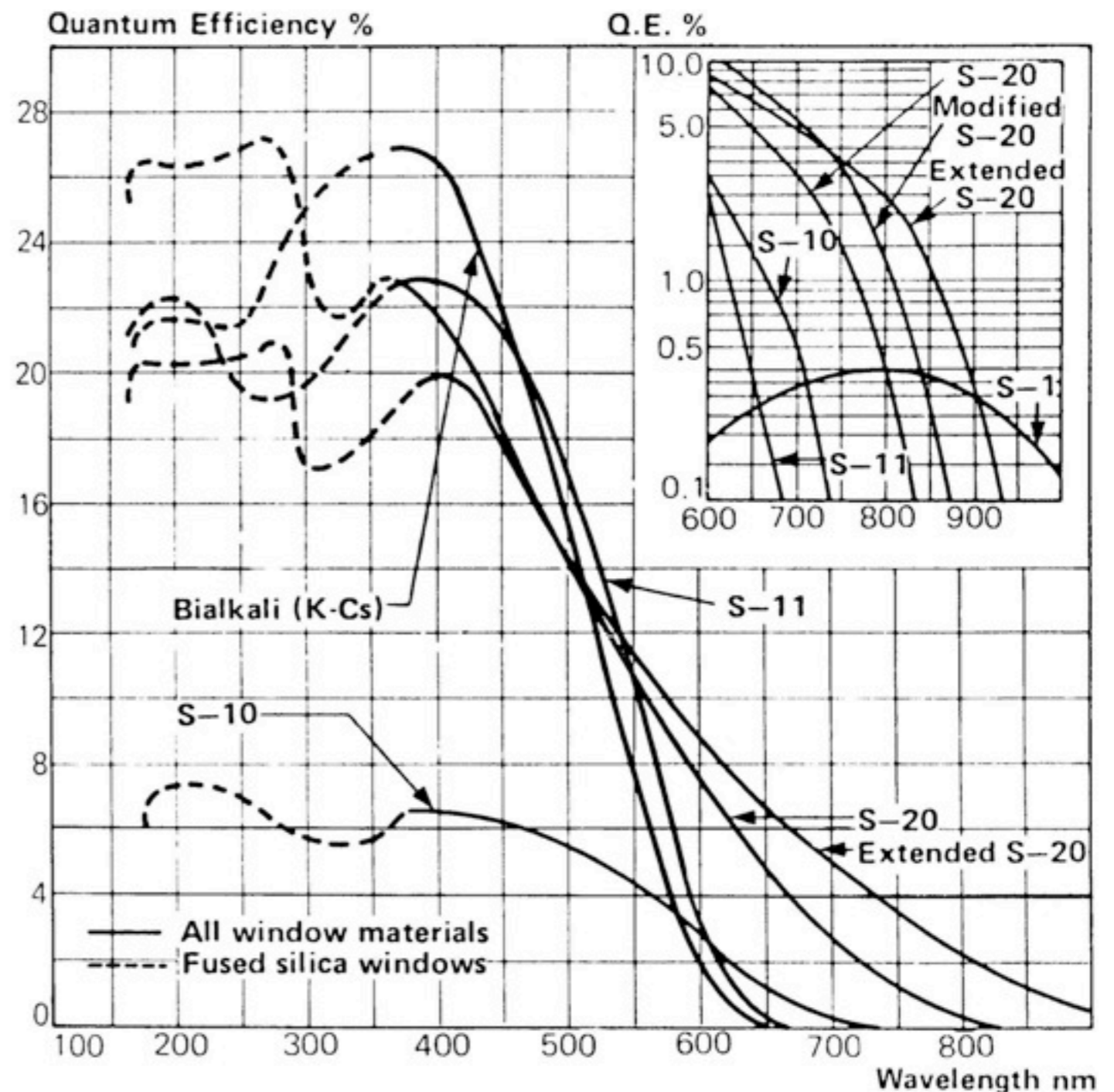


3-step process:

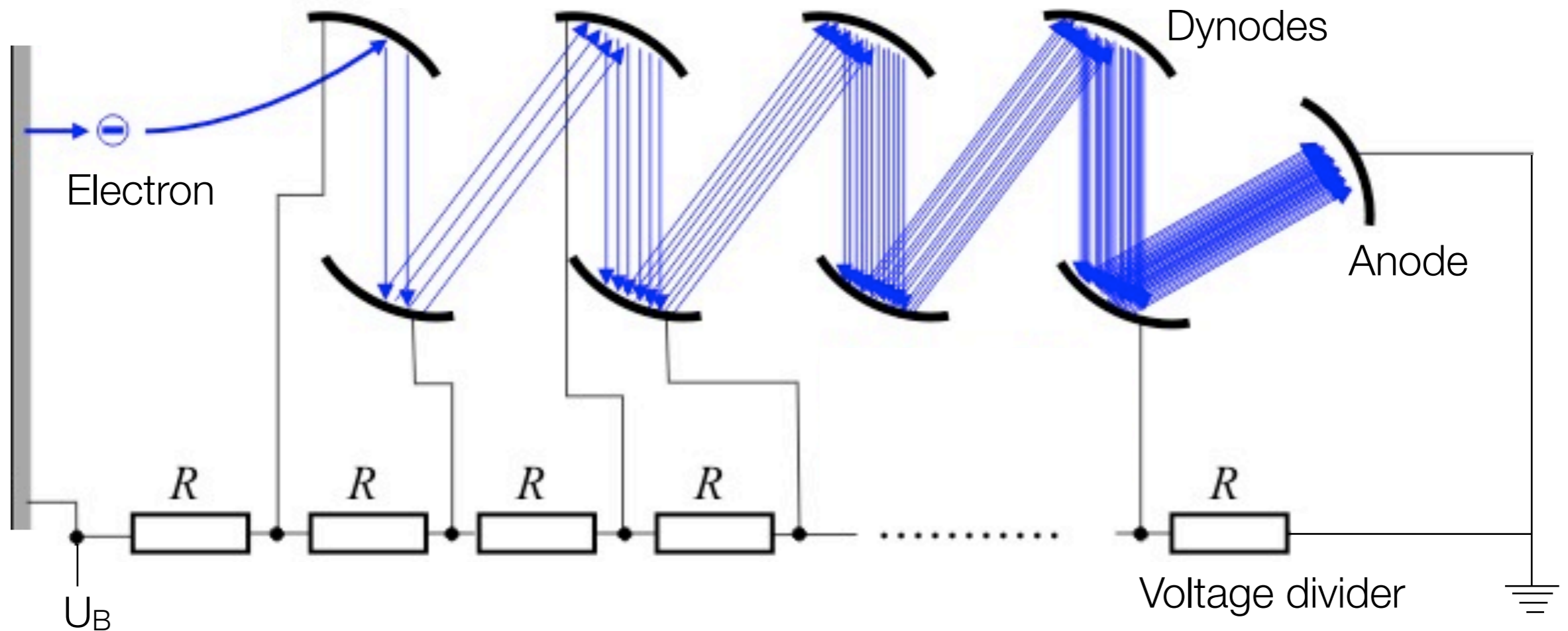
- Electron generation via ionization
- Propagation through cathode
- Escape of electron into vacuum

Q.E. \approx 10-30%

[need specifically developed alloys]



Photomultipliers – Dynode Chain



Multiplication process:

Electrons accelerated toward dynode
 Further electrons produced → avalanche

Secondary emission coefficient:

$$\delta = \#(e^- \text{ produced}) / \#(e^- \text{ incoming})$$

Typical: $\delta = 2 - 10$
 $n = 8 - 15$] $\rightarrow G = \delta^n = 10^6 - 10^8$

Gain fluctuation: $\delta = kU_D$; $G = a_0(kU_D)^n$
 $dG/G = n dU_D/U_D = n dU_B/U_B$

Photomultipliers – Dynode Chain

Optimization of

PMT gain

Anode isolation

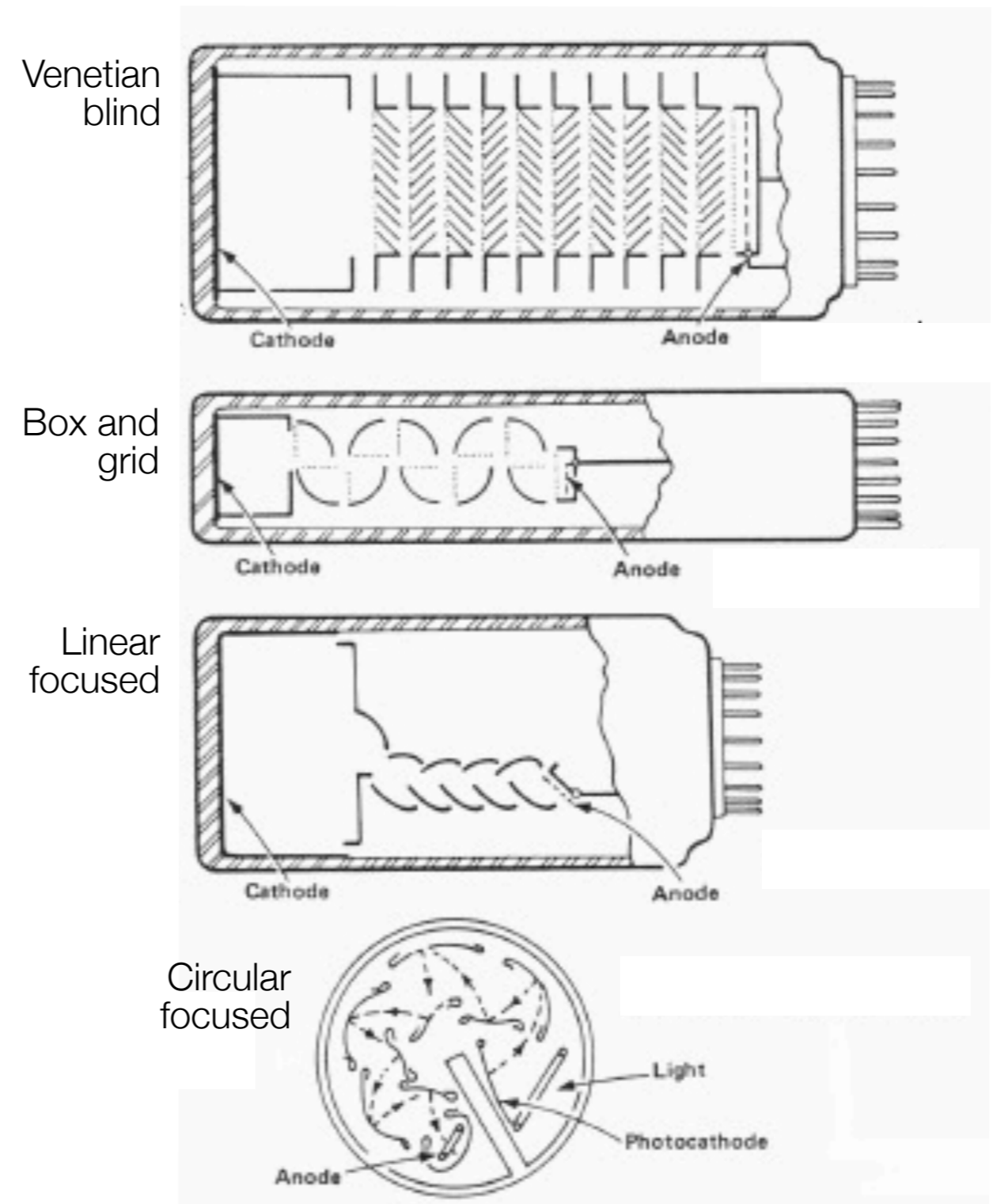
Linearity

Transit time

B-field dependence

PM's are in general
very sensitive to B-fields !

Even to earth field (30-60 μT).
 μ -metal shielding required.



Photomultipliers – Energy Resolution

Energy resolution influenced by:

Linearity of PMT: at high dynode current possible saturation by space charge effects; $I_A \propto n_Y$ for 3 orders of magnitude possible ...

Photoelectron statistics: given by poisson statistics.

$$P_n(n_e) = \frac{n_e^n e^{-n_e}}{n!} \quad \text{with } n_e \text{ given by } dE/dx \dots$$

$$\sigma_n / \langle n \rangle = 1 / \sqrt{n_e}$$

$$n_e = \frac{dE}{dx} \times \frac{\text{Photons}}{\text{MeV}} \times \eta \times \text{Q.E.}$$

For NaI(Tl) and 10 MeV photon;
 photons/MeV = 40000;
 $\eta = 0.2$; Q.E. = 0.25

$n_e = 20000$
 $\sigma_n / \langle n \rangle = 0.7\%$

light collection efficiency

Secondary electron fluctuations:

$$P_n(\delta) = \frac{\delta^n e^{-\delta}}{n!} \quad \text{with dynode gain } \delta; \text{ and with } N \text{ dynodes ...}$$

$$\sigma_n / \langle n \rangle = 1 / \sqrt{\delta}$$

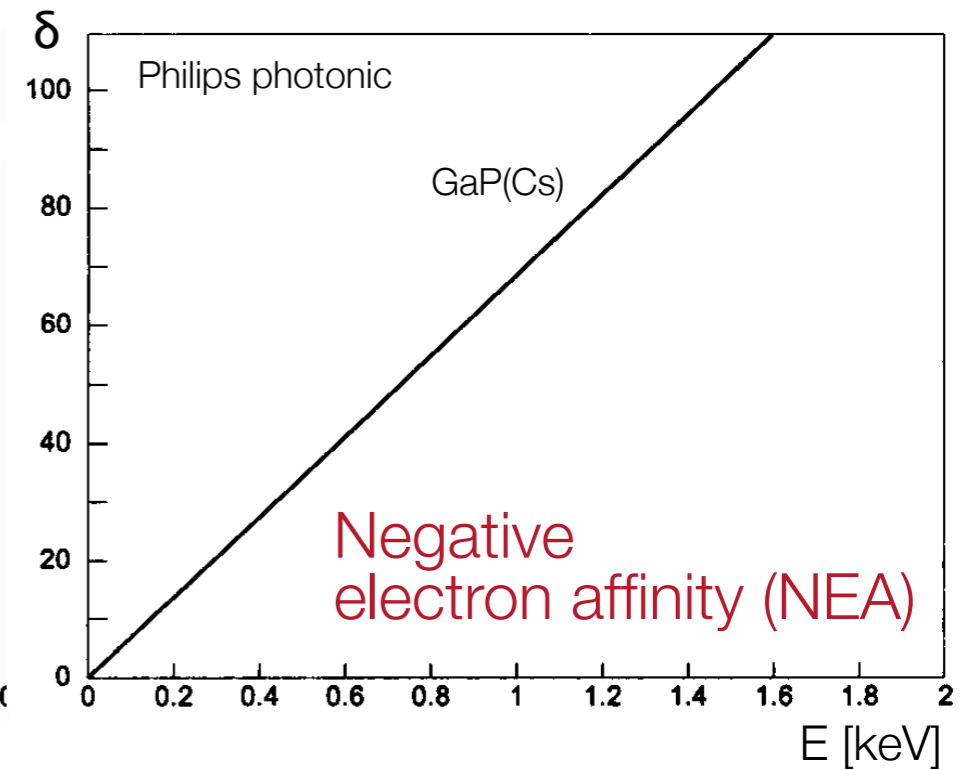
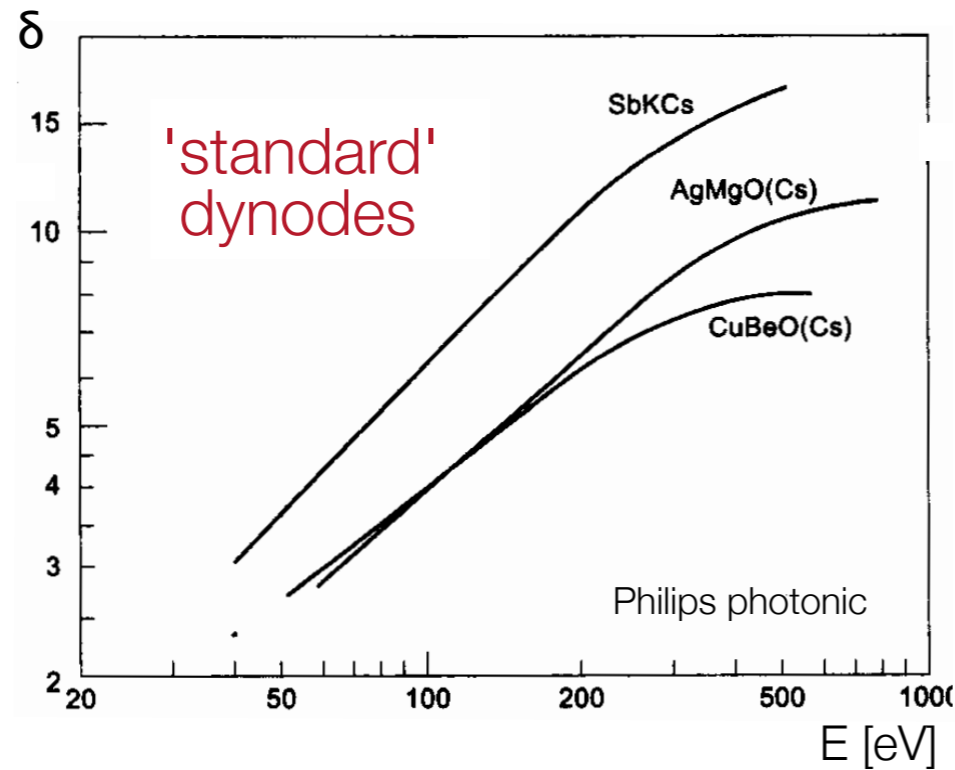
$$\left(\frac{\sigma_n}{\langle n \rangle} \right)^2 = \frac{1}{\delta} + \dots + \frac{1}{\delta^N} \approx \frac{1}{\delta - 1}$$

$\sigma_n / \langle n \rangle$ dominated by first dynode stage ...

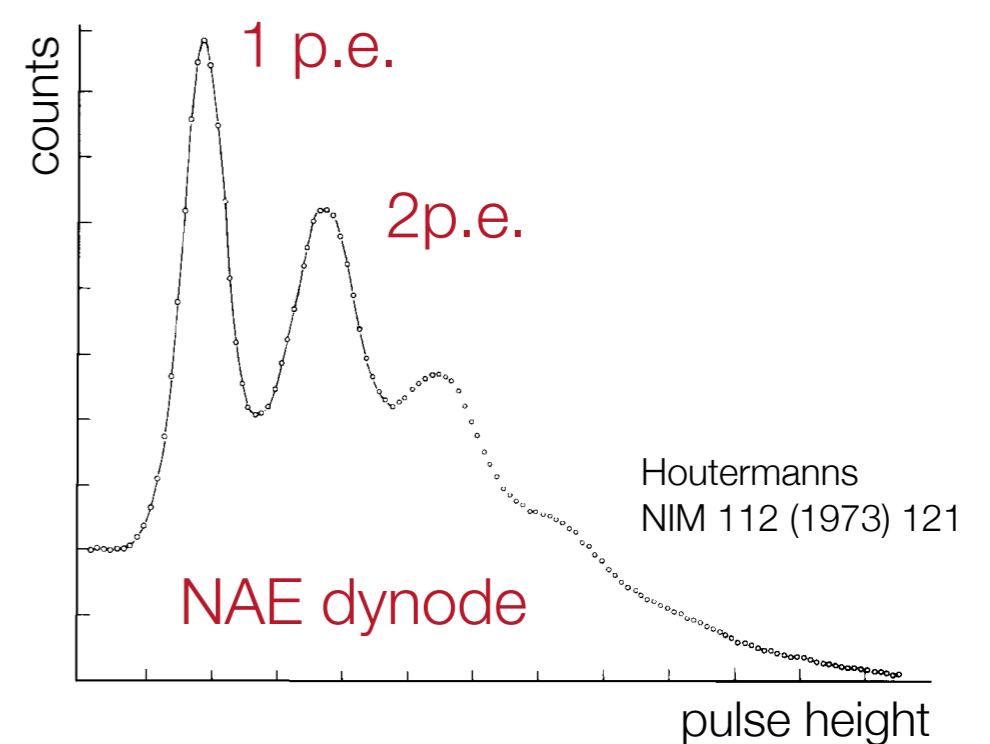
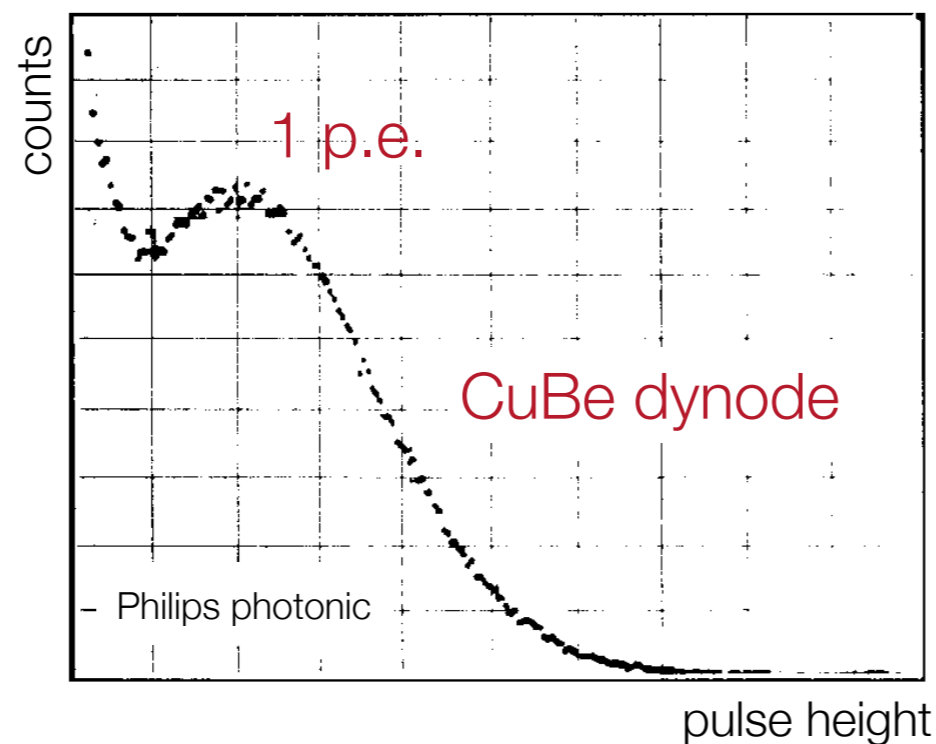
... important for single photon detection

Photomultipliers – Energy Resolution

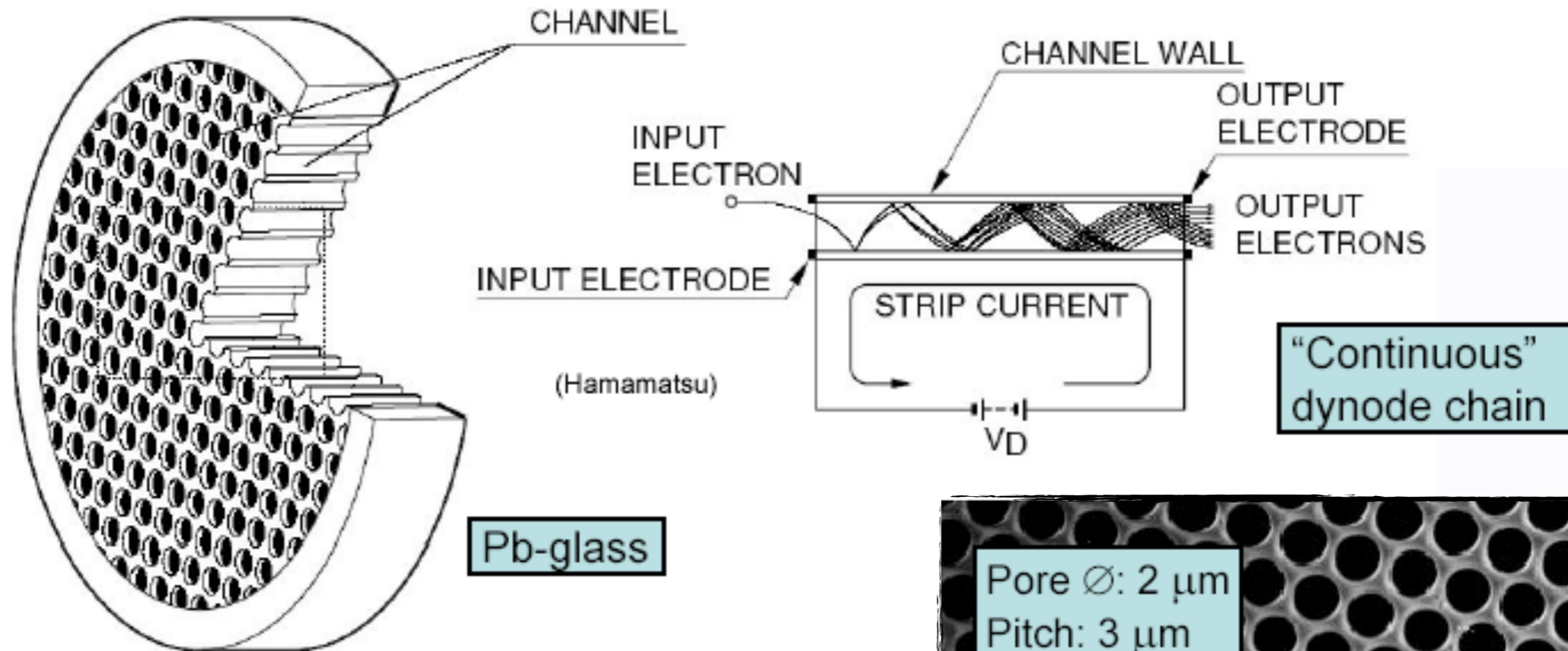
For detection of single photons



Large δ !
... yields better energy resolution



Micro Channel Plate



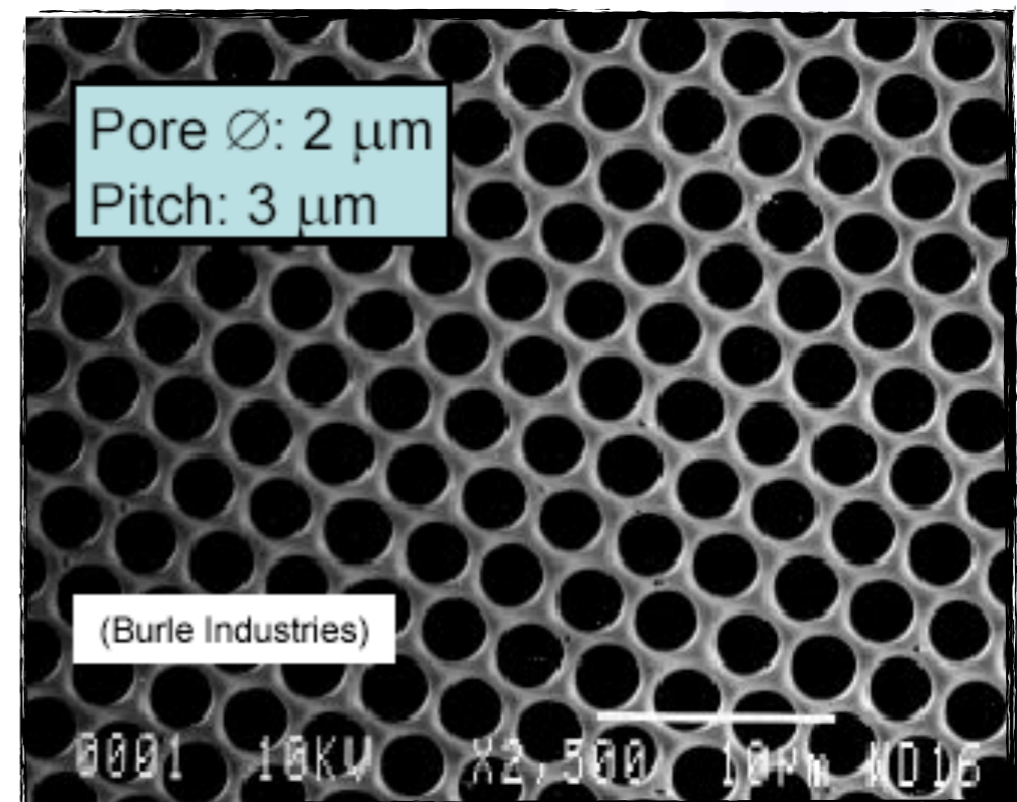
"2D Photomultiplier"

Gain: $5 \cdot 10^4$

Fast signal [time spread ~ 50 ps]

B-Field tolerant [up to 0.1T]

But: limited life time/rate capability



Silicon Photomultipliers

Principle:

Pixelized photo diodes
operated in Geiger Mode

Single pixel works as a binary device

Energy = #photons seen by
summing over all pixels

Features:

Granularity : 10^3 pixels/mm²

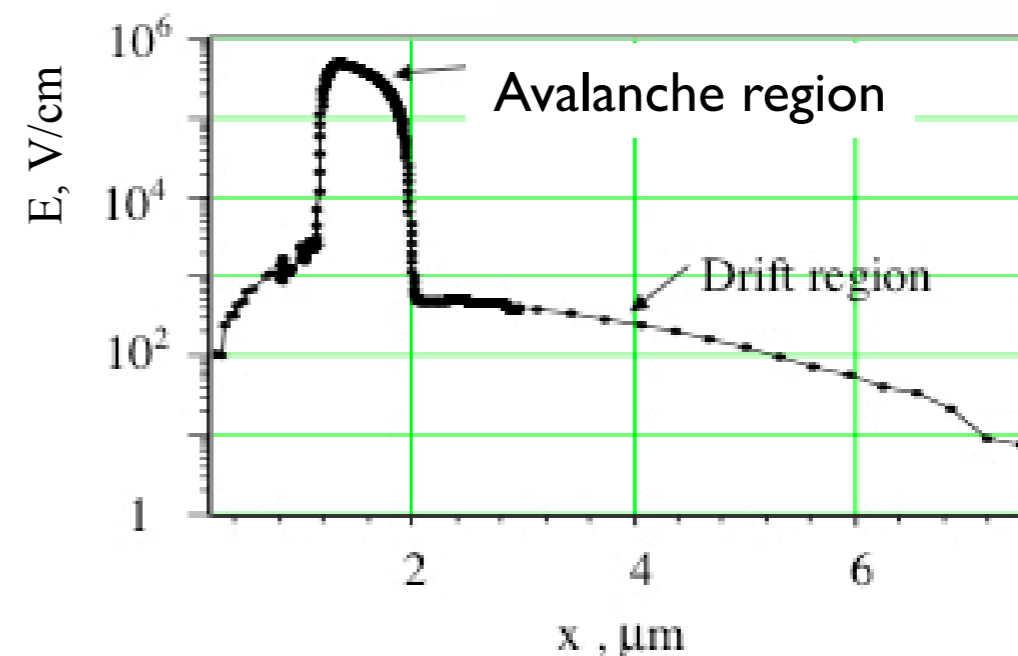
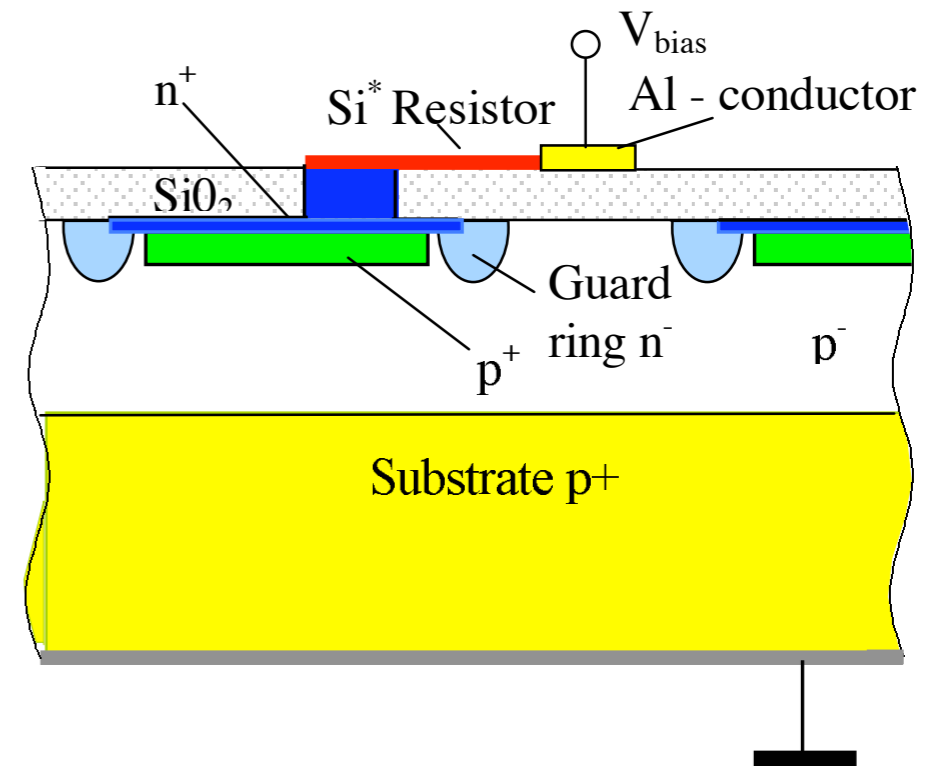
Gain : 10^6

Bias Voltage : < 100 V

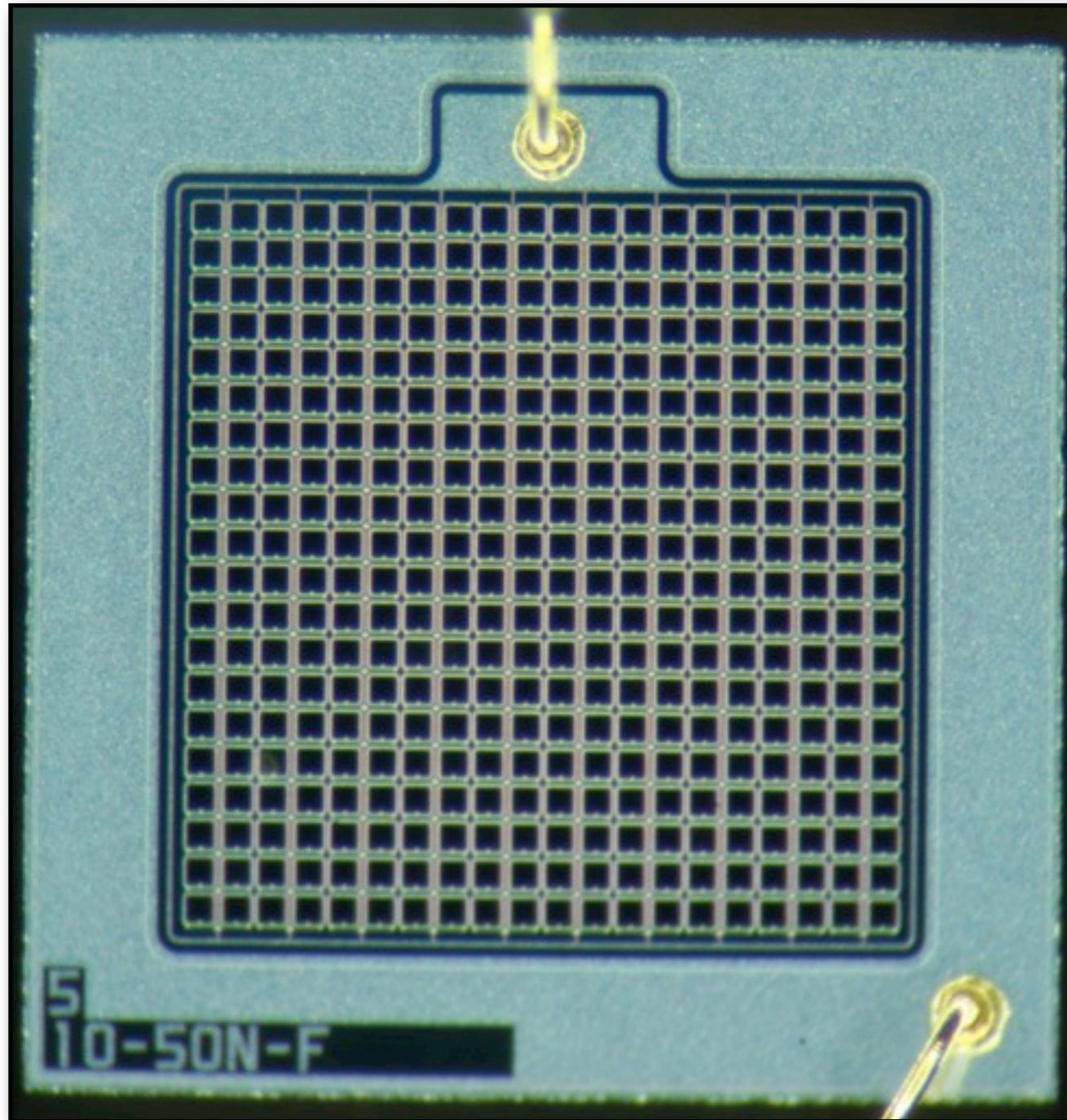
Efficiency : ca. 30 %

Insensitive to magnetic fields!

Works at room temperature ...

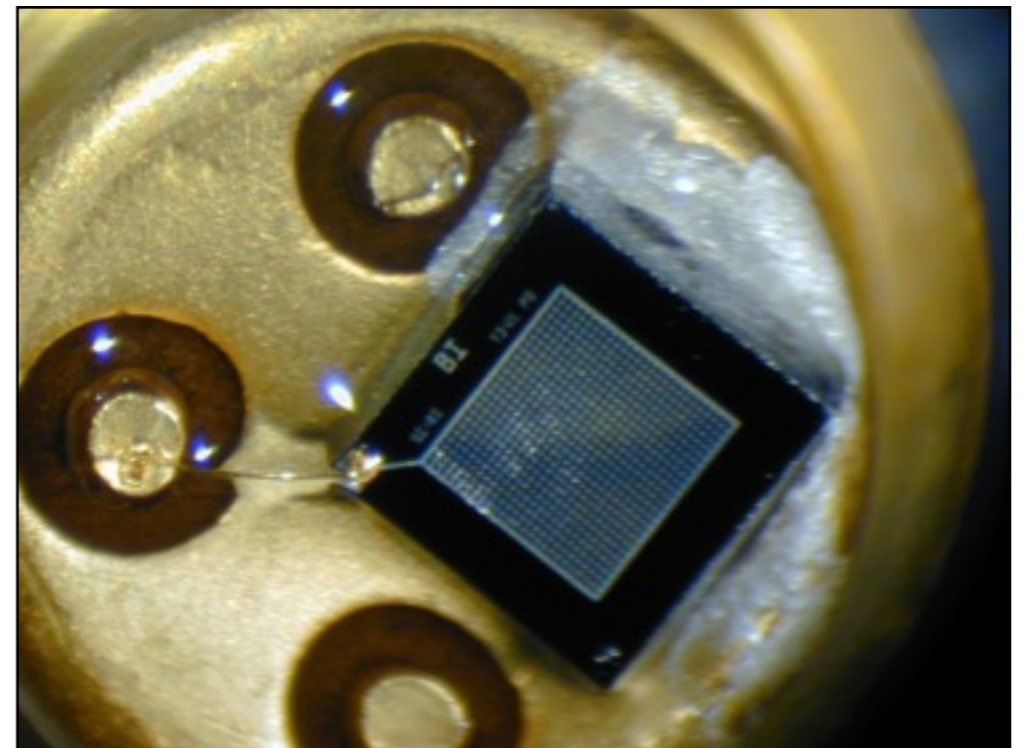


Silicon Photomultipliers

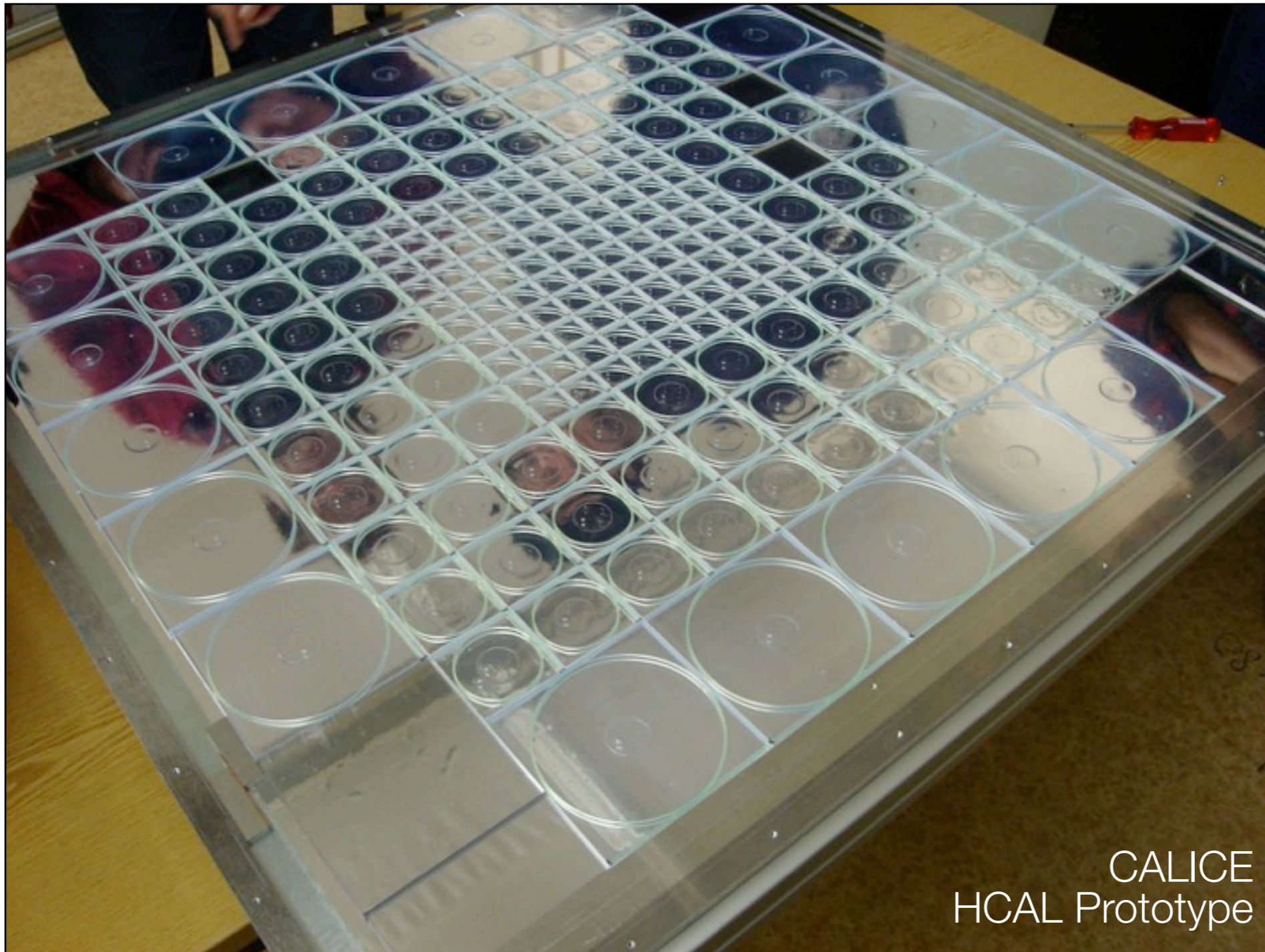


HAMAMATSU
MPPC 400Pixels

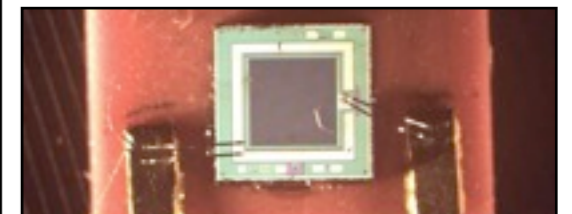
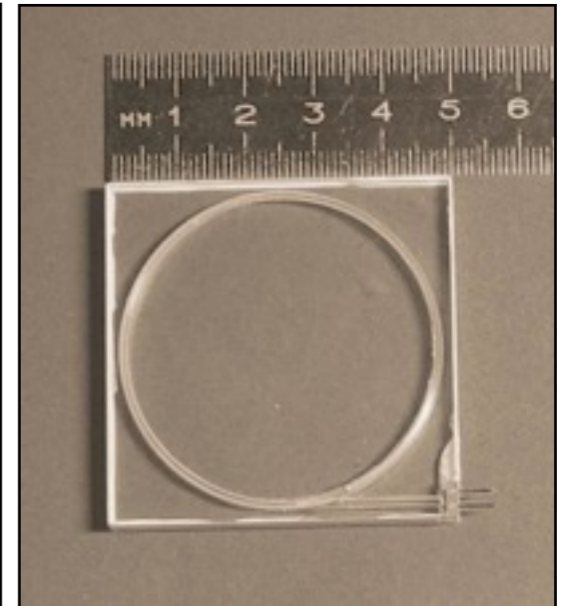
One of the first SiPM
Pulsar, Moscow



Silicon Photomultipliers



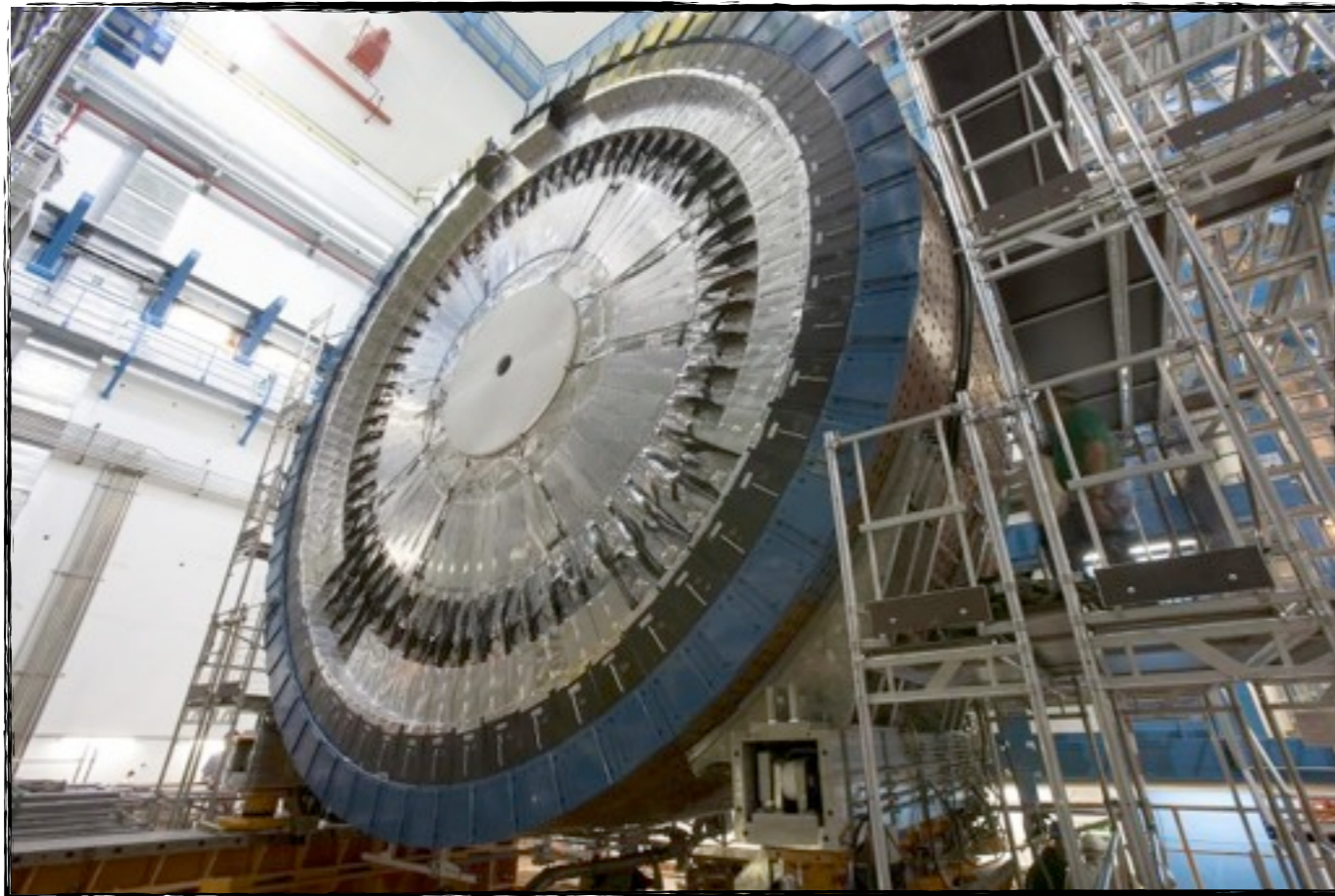
CALICE
HCAL Prototype



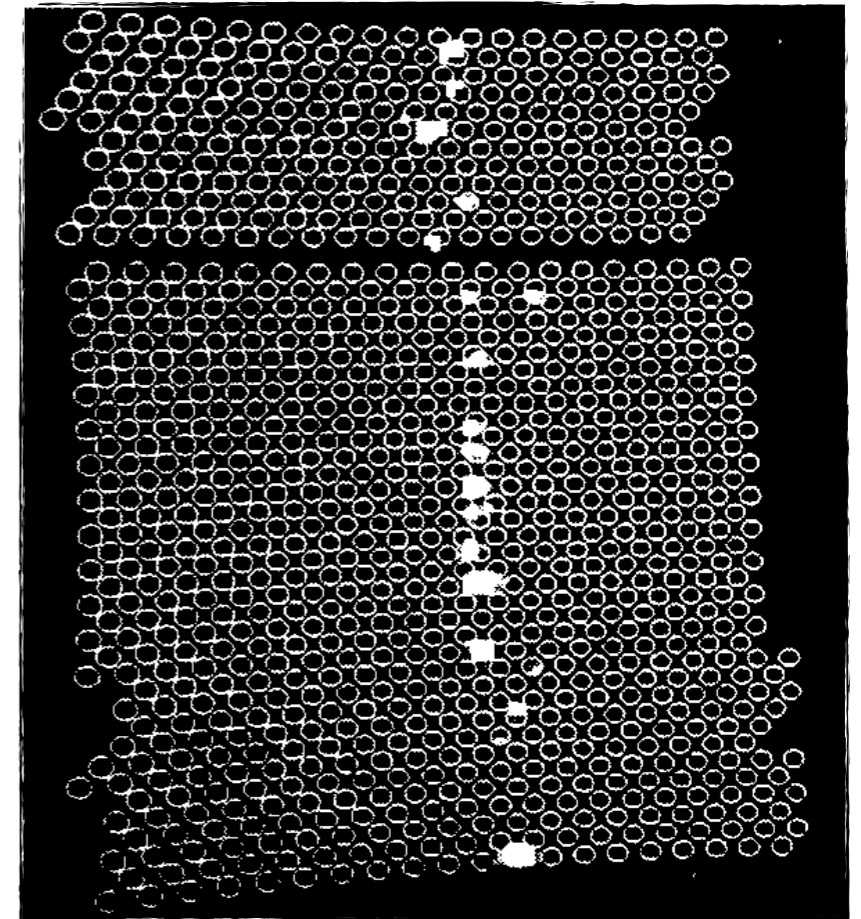
Scintillation Counters – Applications

Time of flight (ToF) counters
Energy measurement (calorimeters)
Hodoscopes; fibre trackers
Trigger systems

ATLAS
Minimum Bias Trigger Scintillators



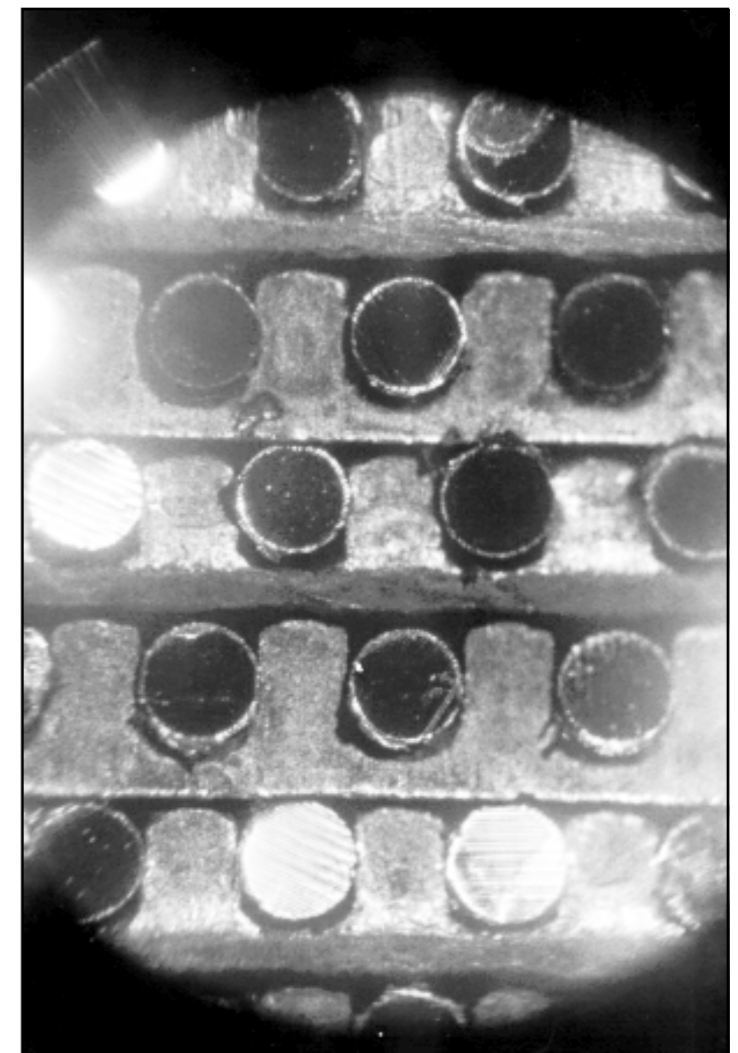
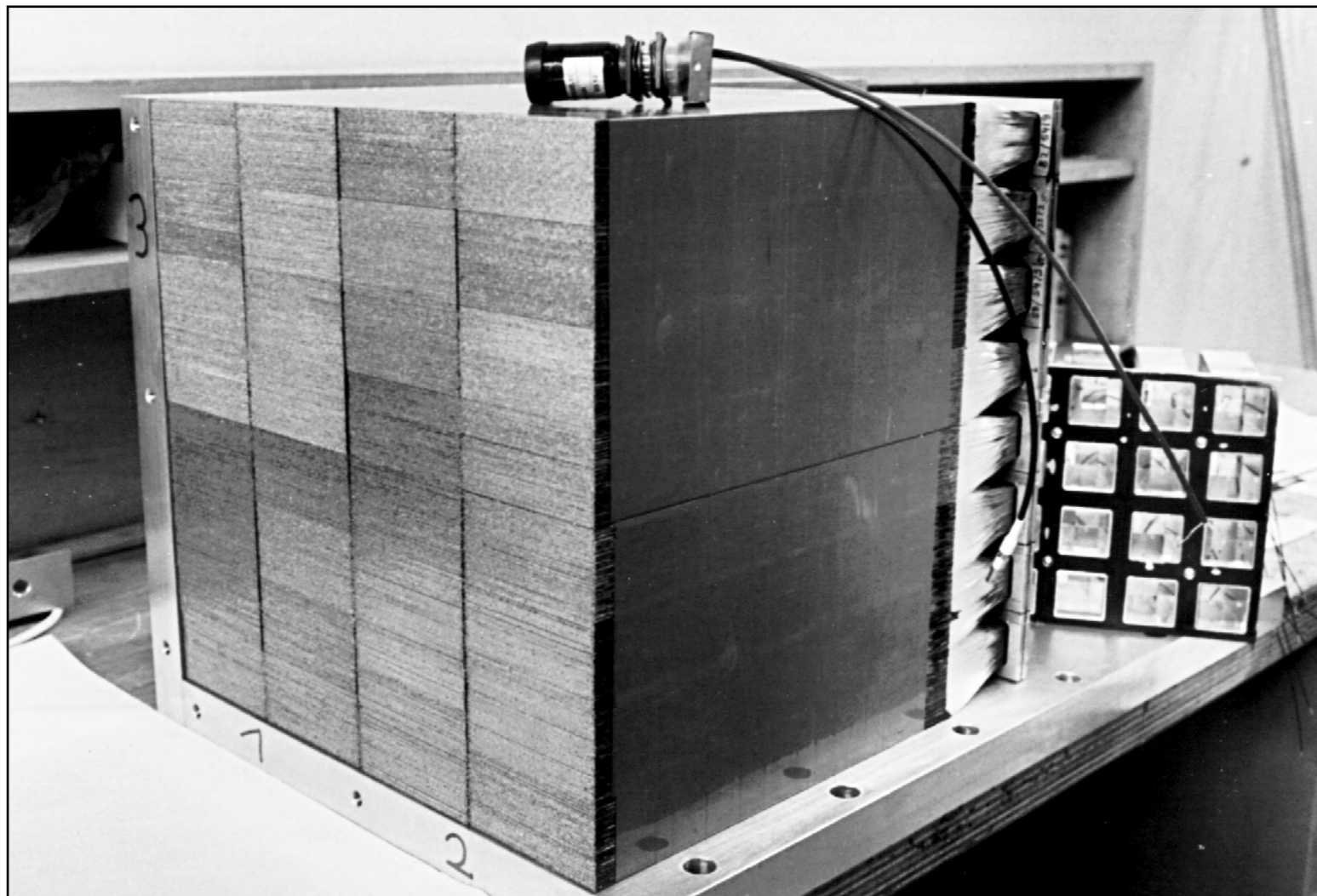
Particle track in
scintillating fibre hodoscope



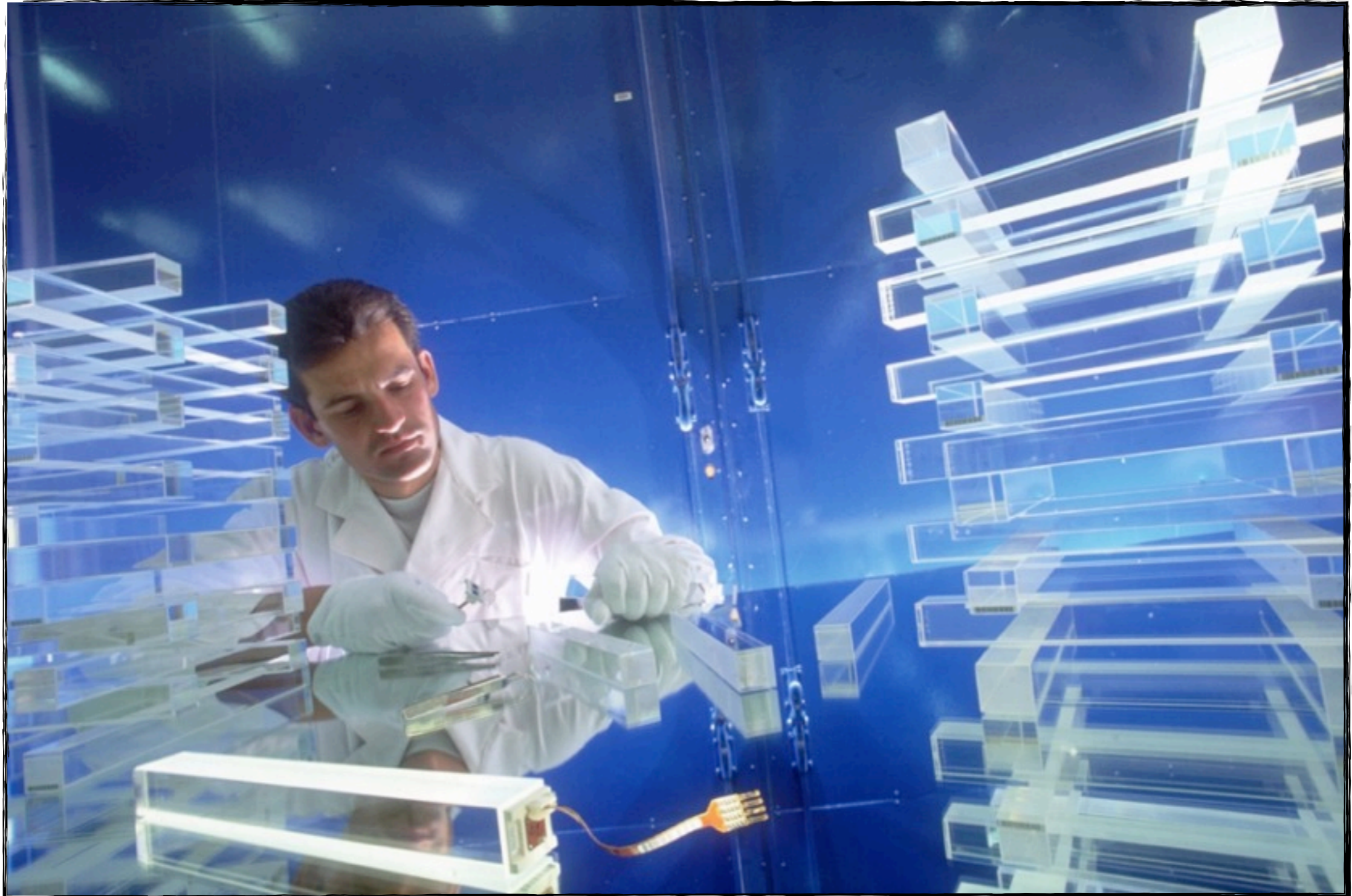
H1 – Spaghetti Calorimeter

Scintillator : BICRON BCF-12

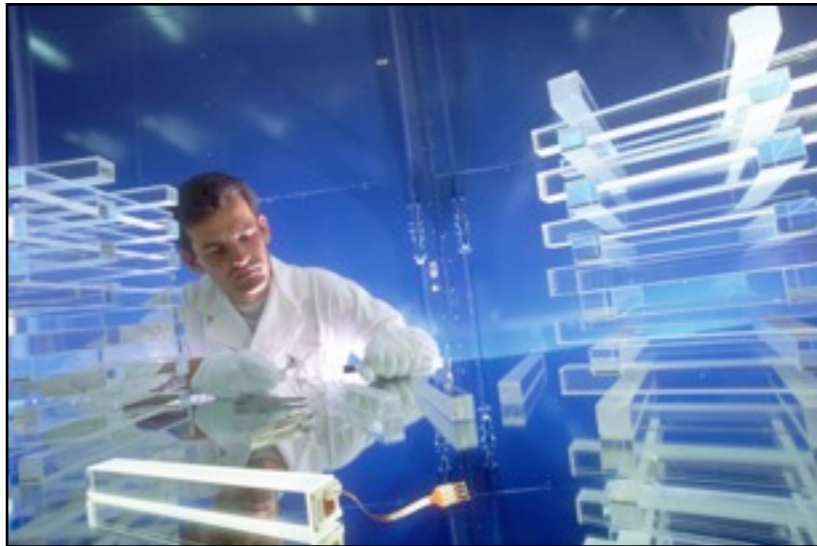
Photosensor : Photomultipliers



CMS – Crystal Calorimeter (ECAL)

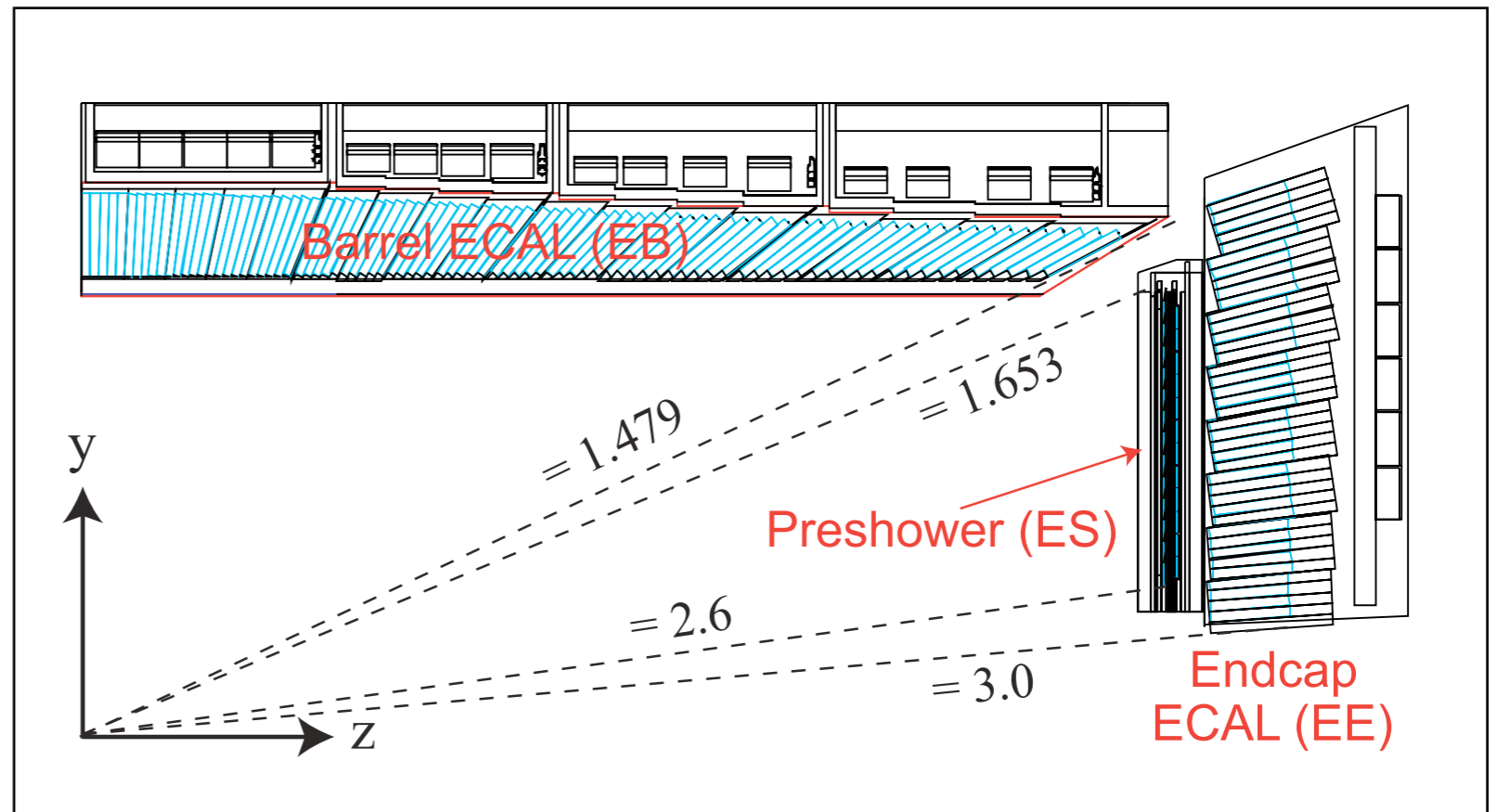
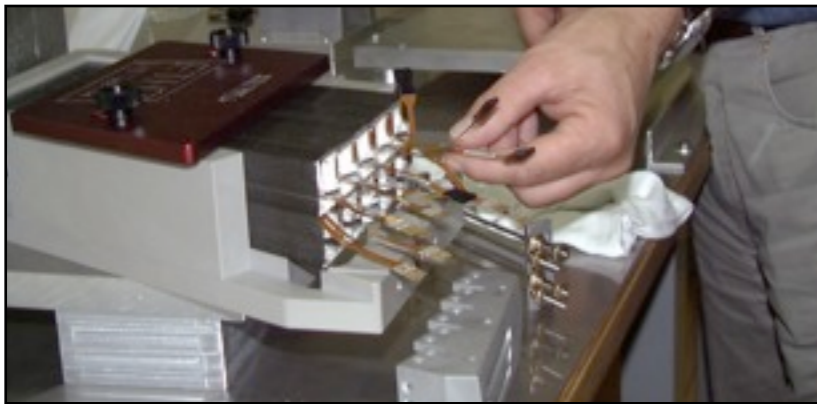


CMS – Crystal Calorimeter (ECAL)

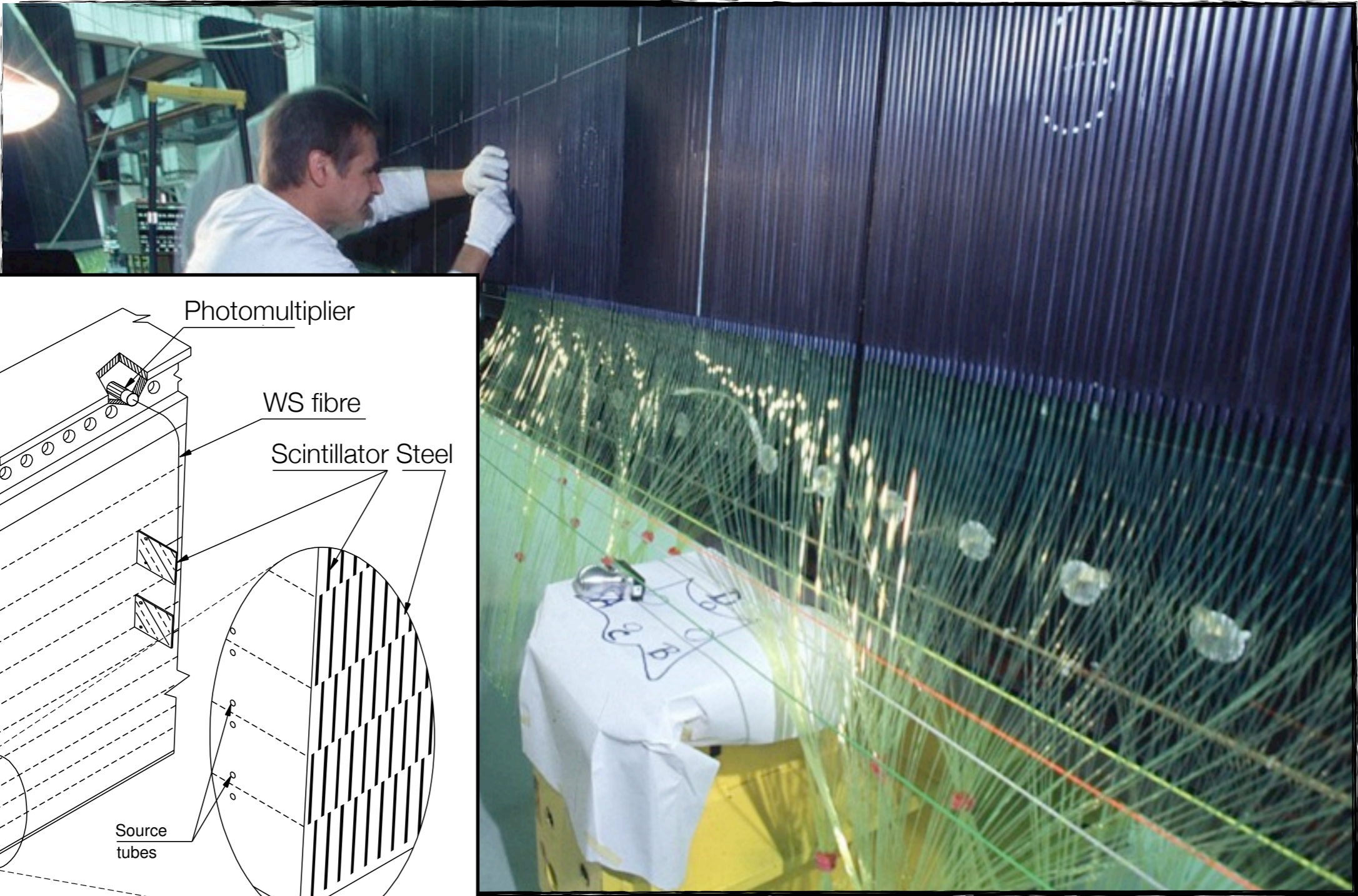


Scintillator : PbWO_4 [Lead Tungsten]
Photosensor : APDs [Avalanche Photodiodes]

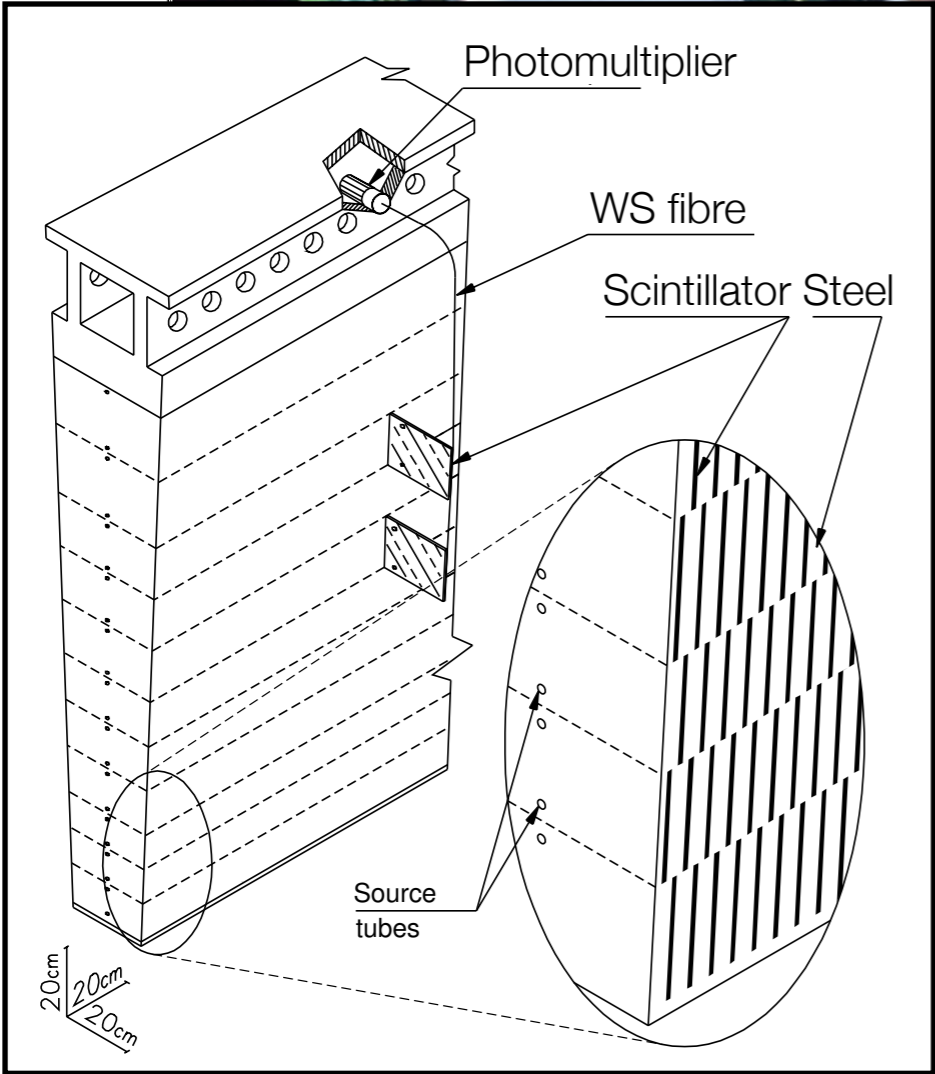
Number of crystals: ~ 70000
Light output: 4.5 photons/MeV



ATLAS – Tile Calorimeter



ATLAS Tile Calorimeter

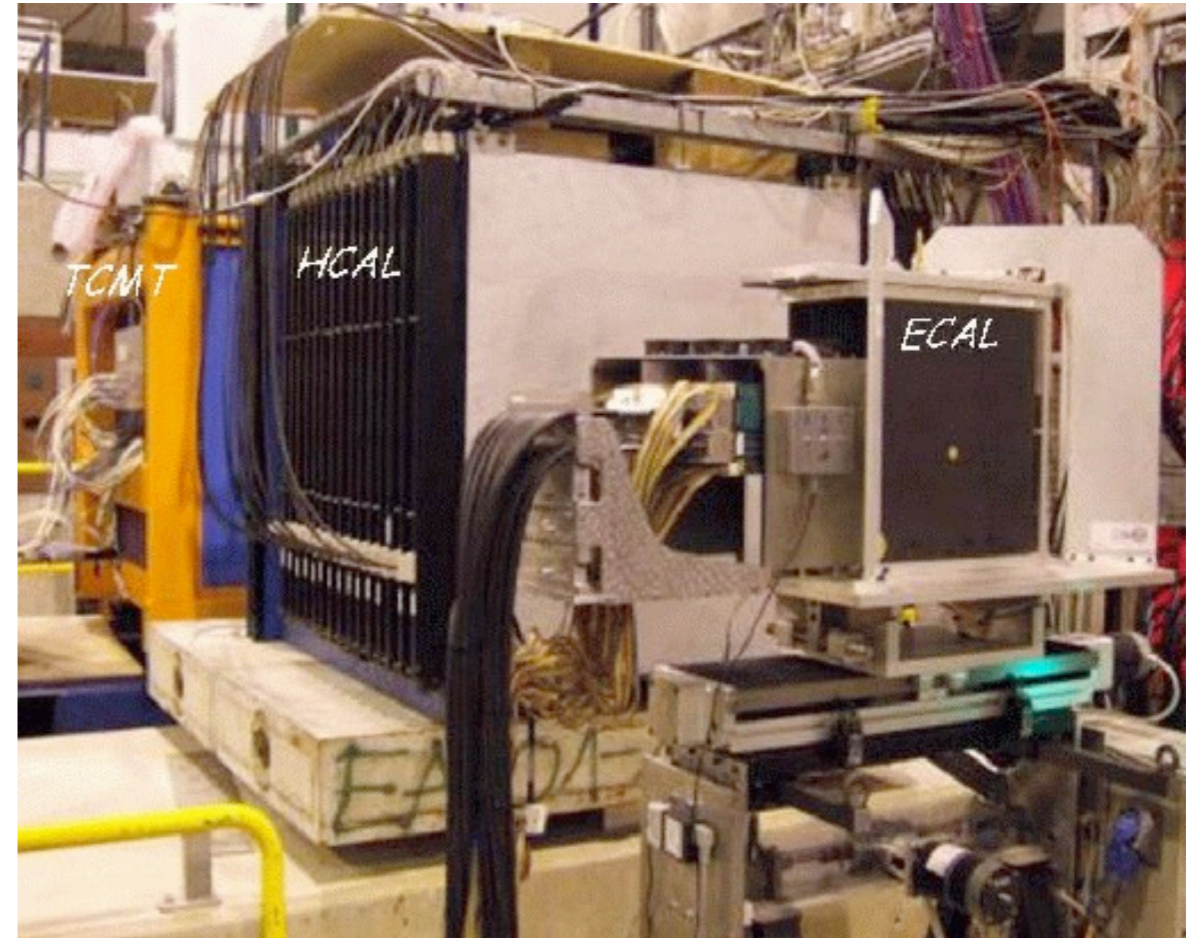
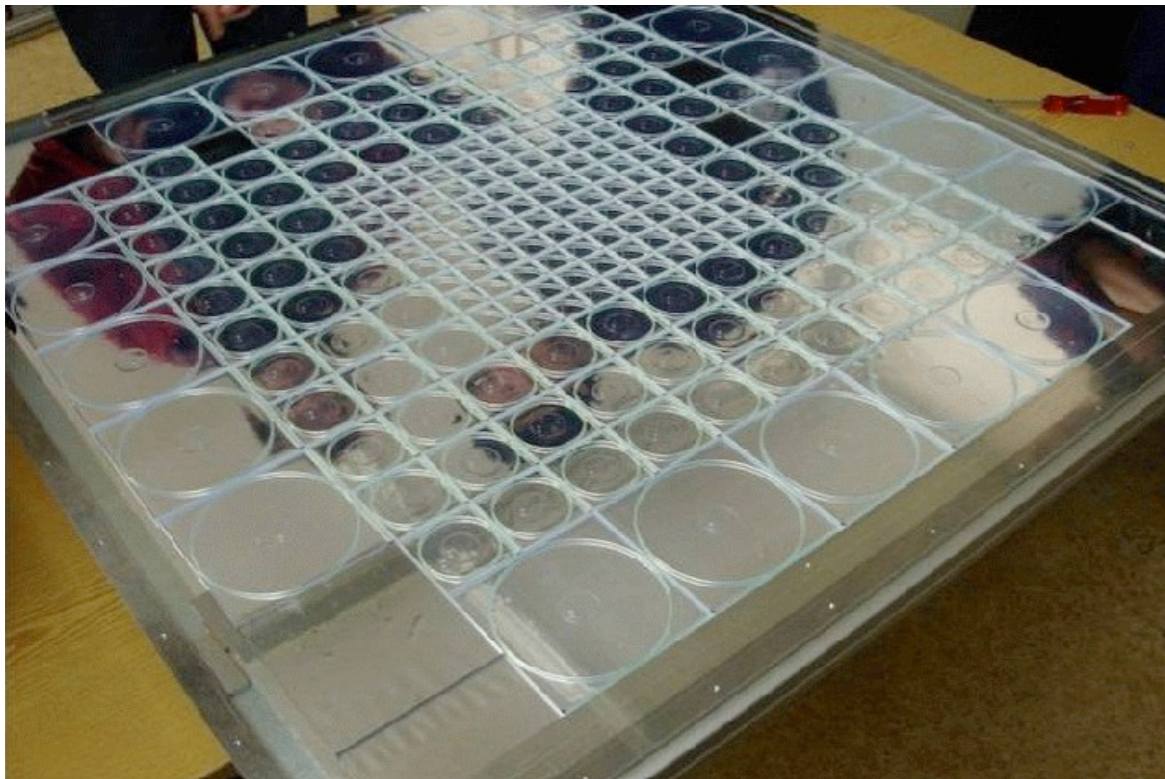


CALICE – Analogue HCAL

1 m³-Prototype
38 layers

Sandwich structure:

- Scintillator Tiles+WLS+SiPMs (.5 cm)
- Stainless steel absorber (1.6 cm)



2006/2007 CERN Testbeam
[2008/09, Fermilab]

Scintillator : Plastic
Photosensor : SiPMs

CALICE – Scintillator ECAL

